

Electromagnetic separators (2)

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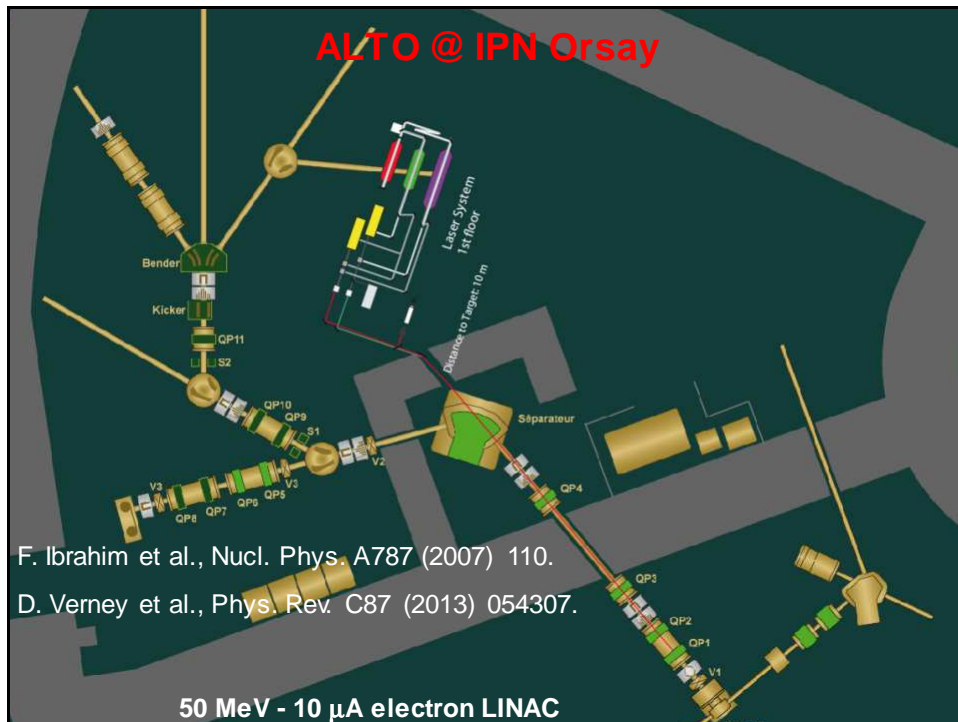
Outline

1. Definitions and history
2. Basics of ion optics and dispersive elements
3. Static fields
 - a) deflection spectrometer
 - b) retardation spectrometer
4. Dynamic fields/separation
 - a) Time-of-Flight spectrometer
 - b) Radiofrequency spectrometer
 - c) Traps
5. Technical realization (ion sources, etc.)
6. “Real examples” for nuclear physics applications
 - a) ISOL
 - b) Recoil separators
 - c) Fragment separators
 - d) Spectrometer

Radioactive ion beam facilities for fission products

Previous, presently operating and future RIB facilities using fission:

$^{252}\text{Cf(sf)}$	CARIBU
$^{235}\text{U}(n_{\text{th}},f)$	OSTIS, OSIRIS, LOHENGRIN, TRIGA-SPEC, CARR-ISOL, PIAFE, MAFF, PIK-ISOL
$^{238}\text{U}(p,f)$	ISOLDE, IRIS, LISOL, JYFL, HRIBF, TRIAC, ISAC-II, SPES, ISOL@MYRRHA, EURISOL
$\text{W}(p,xn..) > ^{238}\text{U}(n,f)$	ISOLDE, IRIS, ISAC-II, EURISOL
$^{12}\text{C}(d,n) > ^{238}\text{U}(n,f)$	PARRNe, SPIRAL-II
$^2\text{H}(d,n) > ^{238}\text{U}(n,f)$	SPIRAL-II
$^9\text{Be}(d,n) > ^{238}\text{U}(n,f)$	PARRNe
$^7\text{Li}(d,n) > ^{238}\text{U}(n,f)$	FRIB
$\text{W}(e^-, \gamma) > ^{238}\text{U}(\gamma, f)$	ALTO, DRIBS, ARIEL
$^1\text{H}, ^9\text{Be}, ^{208}\text{Pb}(^{238}\text{U}, f)$	GSI-FRS, RIKEN, FRIB, FAIR



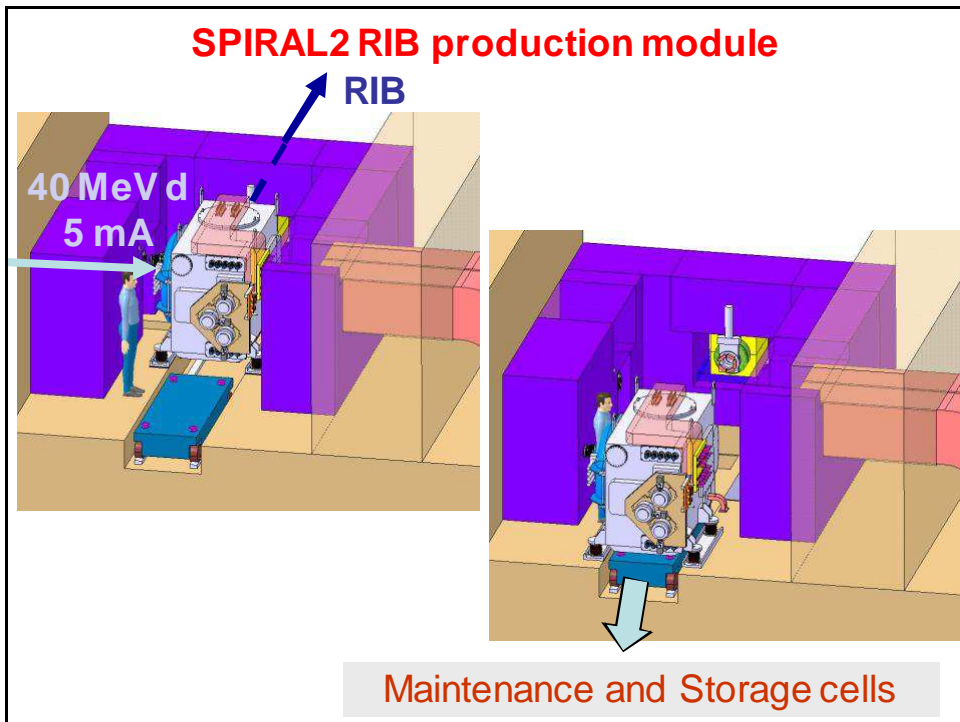
ISAC @ TRIUMF: ARIEL photo-fission upgrade

- 50 MeV, 10 mA electron LINAC
- <100 kW till 2015, 500 kW till 2020
- aim $4.6E13$ fissions/s with liquid Hg converter
- but also 500 MeV protons with maximum $10 \mu\text{A}$ on UC_x

SPIRAL2 facility at GANIL

SP2 Beam time: 44 weeks/y
GANIL Beam time: 35 weeks/y
ISOL RIB Beams: 28-33 weeks/y
GANIL+SP 2 Users: 700-800/y

Cost: 200M€



1-2.2 GeV, multi-MW proton driver

Several direct target stations (ca. 100 kW)

One Hg spallation + fission target station (>1 MW, i.e. 1E15 fissions/s)

Multiple user operation in parallel

Low-energy beam area

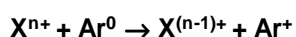
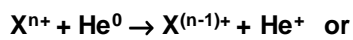
Post-acceleration with LINAC up to ca. 10 A.MeV

Post-acceleration to ca. 100 A.MeV with LINAC or cyclotron

Fragmentation of post-accelerated RIBs

Commissioning: > 2020?

IGISOL method

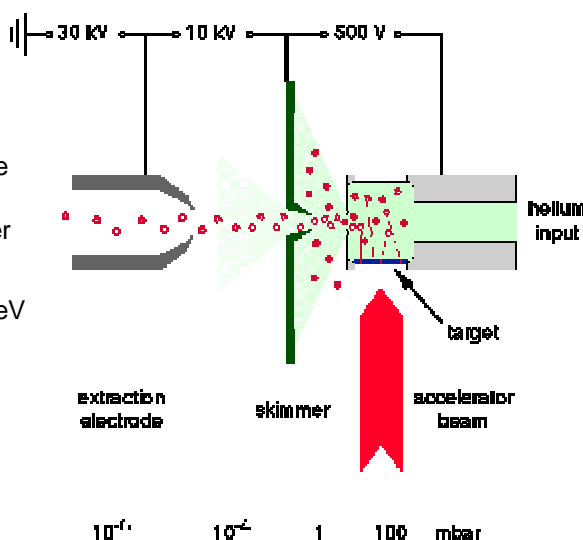


rapid reduction of ionic charge state to 2+ or 1+ by charge exchange reactions with buffer gas

IP(He)=24.6 eV, IP(Ar)=15.8 eV



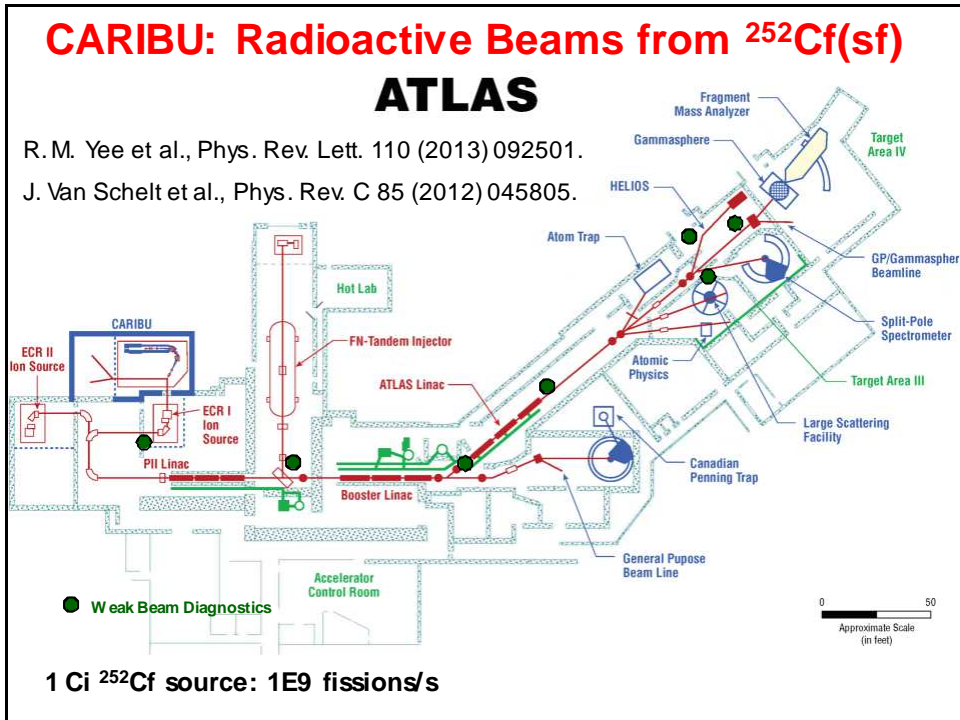
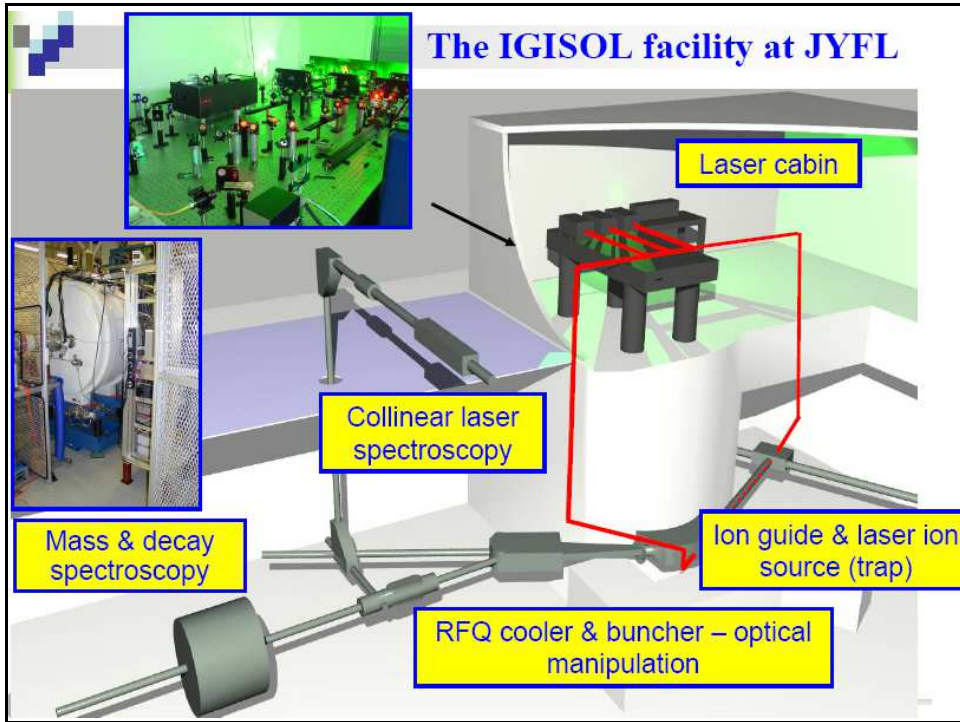
remains in 1+ or 2+ charge state until charge exchange reaction with impurity molecule (O₂, N₂,...) occurs



Volatility of the elements

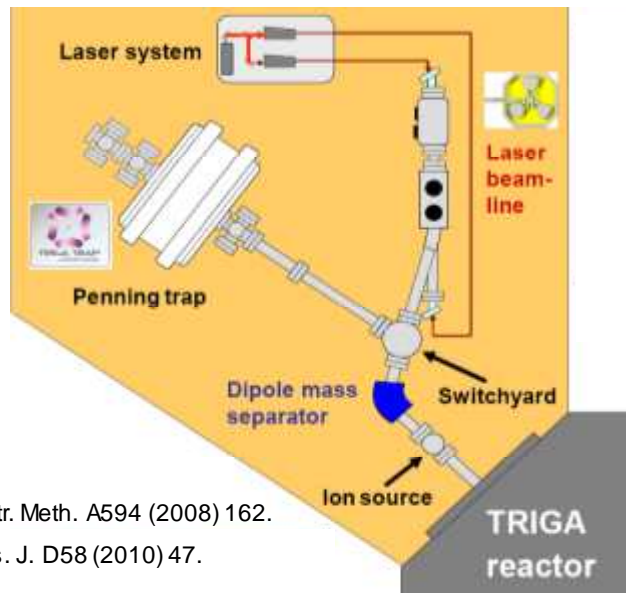
1	T (p vapor > 0.01 mbar) < 100 °C																2				
H	T (p vapor > 0.01 mbar) < 400 °C																He				
3	T (p vapor > 0.01 mbar) < 1000 °C														5	6	7	8	9	10	
Li	Be	T (p vapor > 0.01 mbar) < 2000 °C														B	C	N	O	F	Ne
11	T (p vapor > 0.01 mbar) > 2000 °C														13	14	15	16	17	18	
Na	Mg															Al	Si	P	S	Cl	Ar
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36				
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr				
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54				
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe				
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86				
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn				
87	88	89	104	105	106	107	108	109	110	111	112	113	114	115	116	117	118				
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt	Ds	Rg	Cn	Fl	Lv	Ts	Og	119	120				

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr



TRIGA-SPEC at Mainz reactor

- 0.5 mg ^{235}U or 0.5 mg ^{239}Pu or 0.3 mg ^{249}Cf
- $1.8\text{E}11$ n./cm²/s
- $2\text{E}8$ fissions/s



J. Ketelaer et al., Nucl. Instr. Meth. A594 (2008) 162.

J. Ketelaer et al., Eur. Phys. J. D58 (2010) 47.

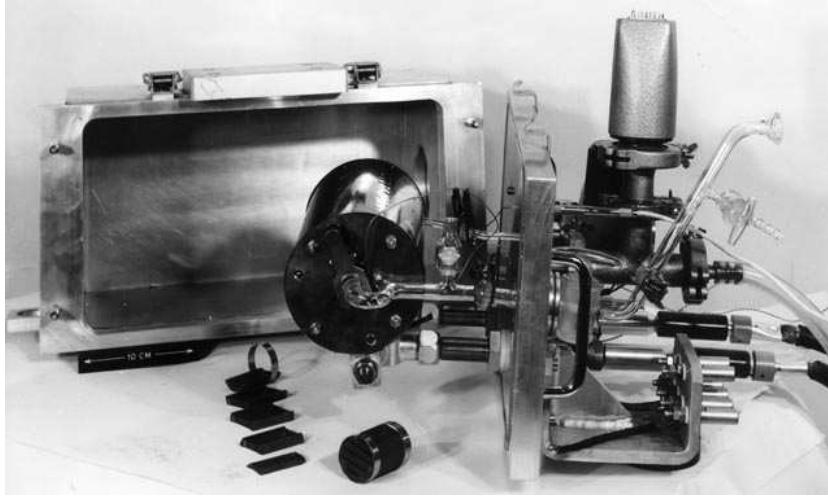
Optimize RIB intensity

All steps of the separation chain need to be optimized!

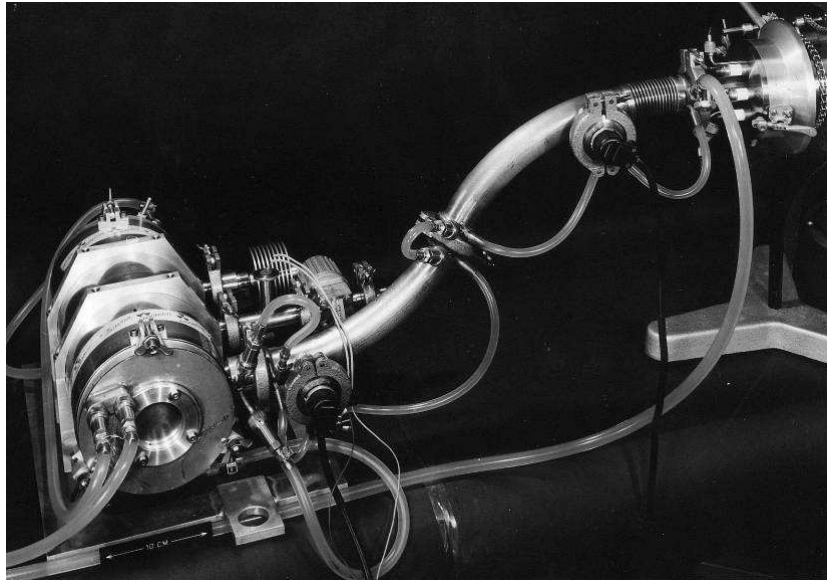
$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$



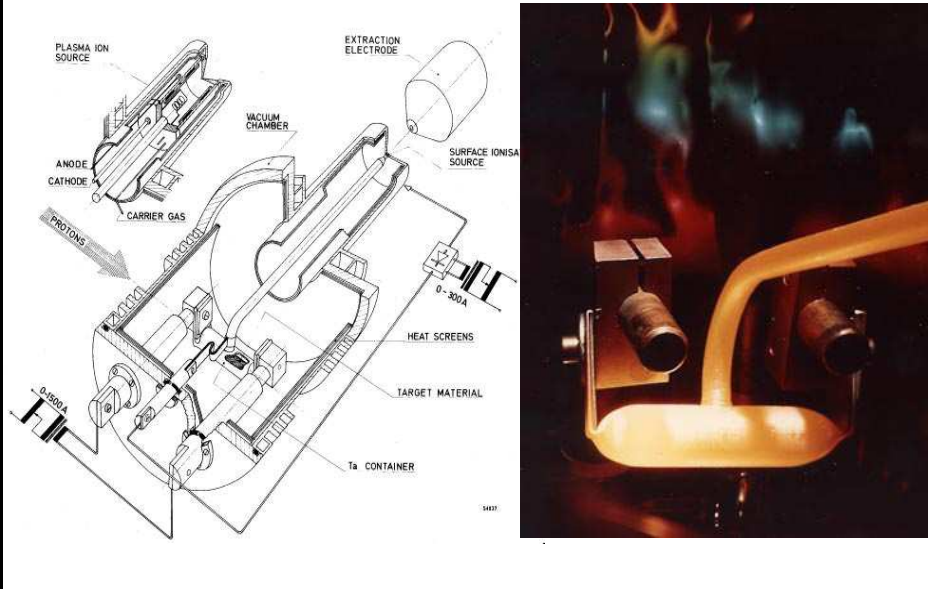
ISOLDE Target (1967)



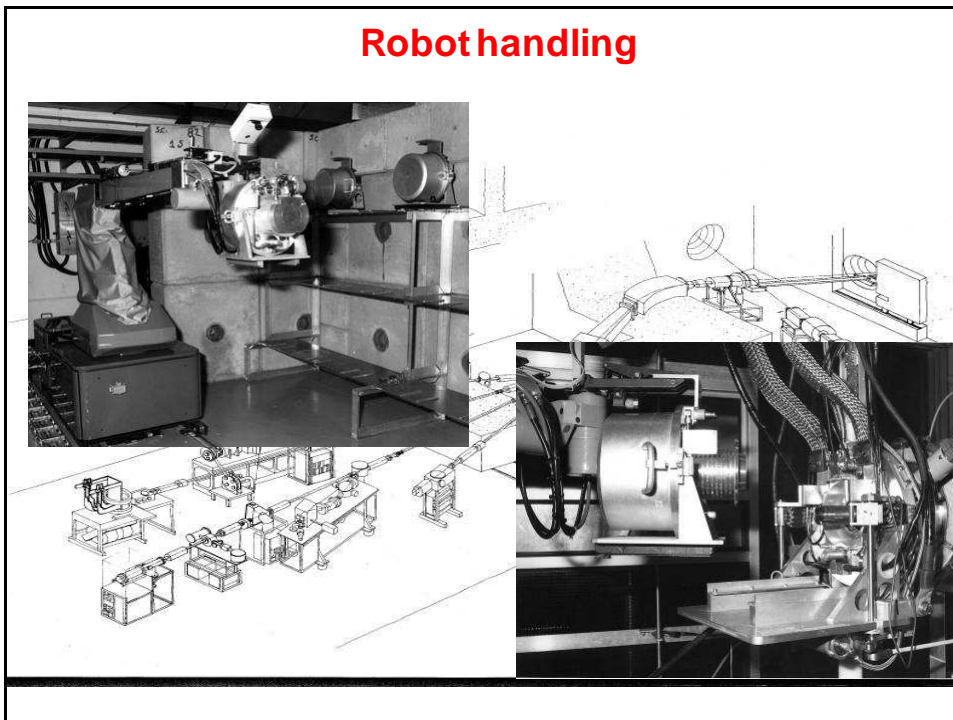
ISOLDE Target and ion source (1968)



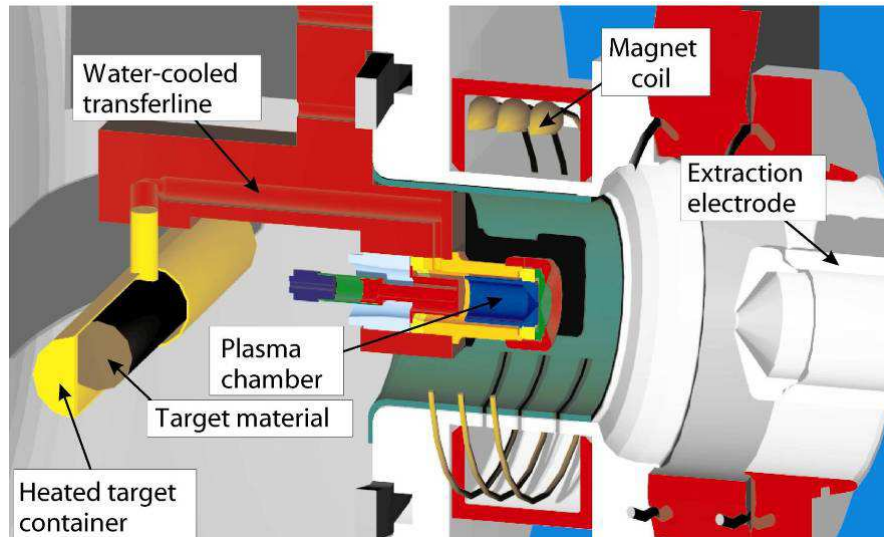
ISOLDE Compact target and ion source (1974)



Robot handling



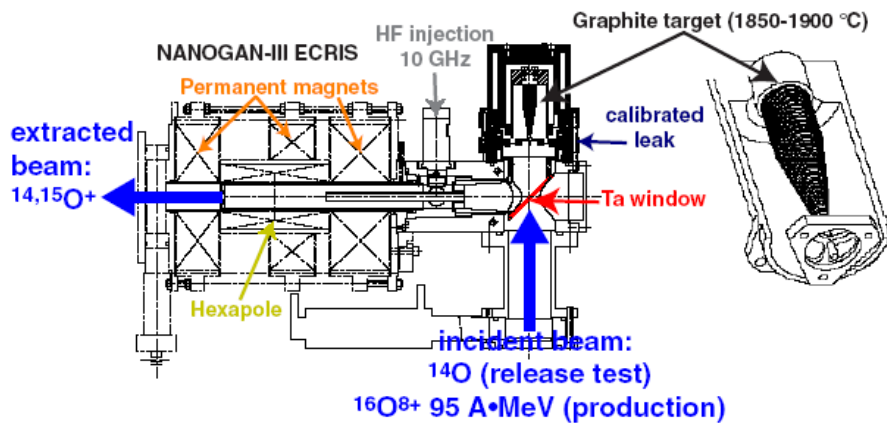
ISOLDE target and ion source unit



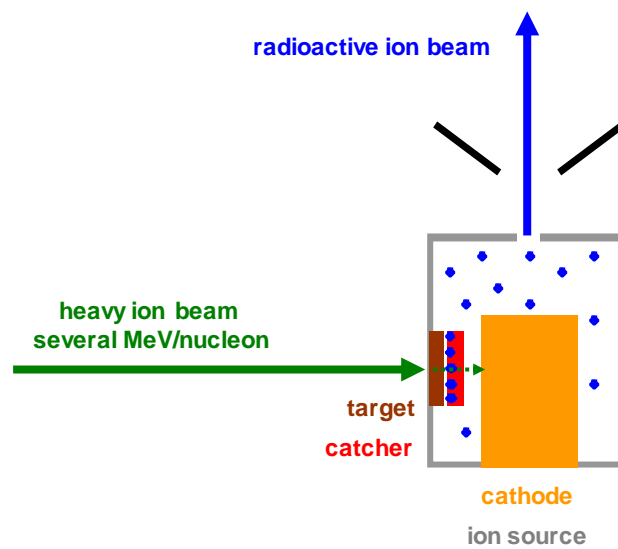
Historical development

- Miniaturisation** ⇒ faster release
- Standardisation** ⇒ easier mass-production
- Remote handling** ⇒ higher activities

SPIRAL target and ion source unit



GSI-ISOL target and ion source unit



Variants of ISOL facilities

- 1a **protons on thick (heavy) target:** fragmentation, spallation, fission
ISOLDE-CERN (1.4 GeV), IRIS-PNPI (1 GeV), ISAC-TRIUMF (0.5 GeV)
- 1b **direct reactions in thick target**
CRC Louvain-la-Neuve, HRIBF Oak Ridge, TRIAC Tokai
- 1c **fission in thick target**
OSIRIS (Studsvik), HRIBF Oak Ridge, TRIAC Tokai, SPIRAL2 (GANIL)
- 2 **projectile fragmentation in thick (carbon) target**
SPIRAL (GANIL), DRIBS (Dubna), EXCYT (LNS Catania)
- 3 **fusion-evap. or multinucleon transfer in thin target plus solid catcher**
GSI-ISOL, UNIRIB (ORNL), DOLIS (Daresbury), LISOL (Leuven), IMP Lanzhou, TRIuP KVI Groningen, MASHA (Dubna), SPIRAL2 (GANIL)
- 4 **fusion-evap., direct reaction or fission in thin target plus gas catcher (Ion Guide ISOL = IGISOL)**
IGISOL (Jyväskylä), LISOL (Leuven), ...
- 5 **liquid helium catcher**
JYFL Jyväskylä, KVI Groningen

ISOL targets

Target materials:

1. **molten metals:** Ge, Sn, La, Pb, Bi, U,...
2. **solid metals:** Ti, Zr, Nb, Mo, Ta, W, Th,...
3. **carbides:** Al₄C₃, SiC, VC, ZrC, LaC_x, ThC_x, UC_x,...
4. **oxides:** MgO, Al₂O₃, CaO, TiO_x, ZrO₂, CeO_x, ThO₂,...
5. **others:** graphite, borides, silicides, sulfides, zeolithes,...

Target dimensions:

target container: 20 cm long, 2 cm diameter

target thickness 2—200 g/cm², 10—100% of bulk density

micro-dimensions of foils, fibers or pressed powder: 1—30 μm

Radiochimica Acta 89 (2001) 749.

Diffusion characteristics

Bad diffusion hosts (narrow and/or stiff crystal lattice):

Re, diamond, SiC,...

Good diffusion hosts (wide crystal lattice):

Ti, Zr, Hf (fcc metals), Nb, Ta, graphite,
polycrystalline oxides (in particular fibers!)

Characteristic diffusion length:

$$d = (2 n D t)^{1/2} \quad n=1 \text{ (foil)}, n=2 \text{ (fiber)}, n=3 \text{ (sphere)}$$

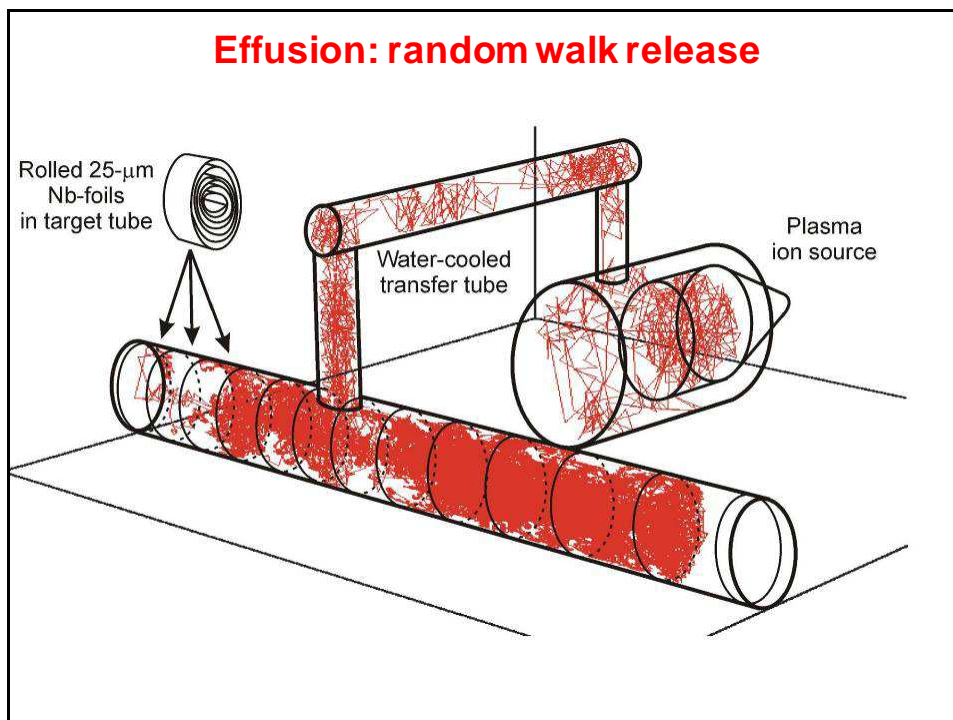
Maximize D and minimize diffusion path:

⇒ thin metal foils (2 μm ... 30 μm)

⇒ fine powders (μm)

⇒ thin fibers (some μm)

Effusion: random walk release



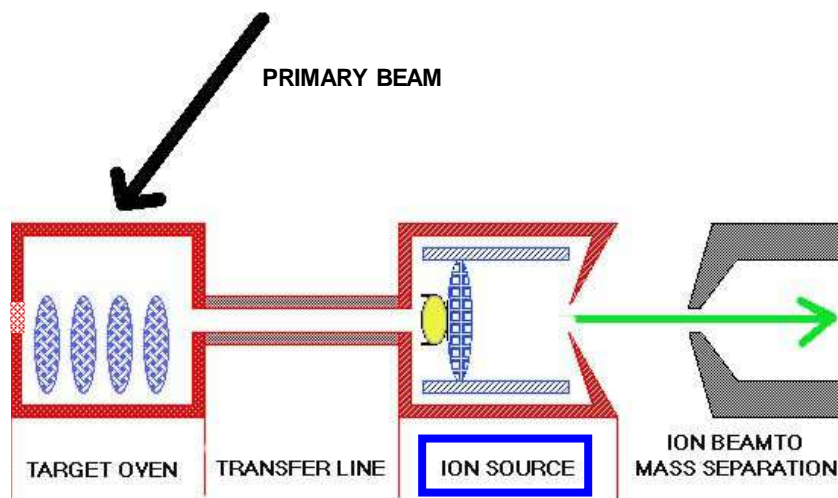
Optimize RIB intensity

All steps of the separation chain need to be optimized!

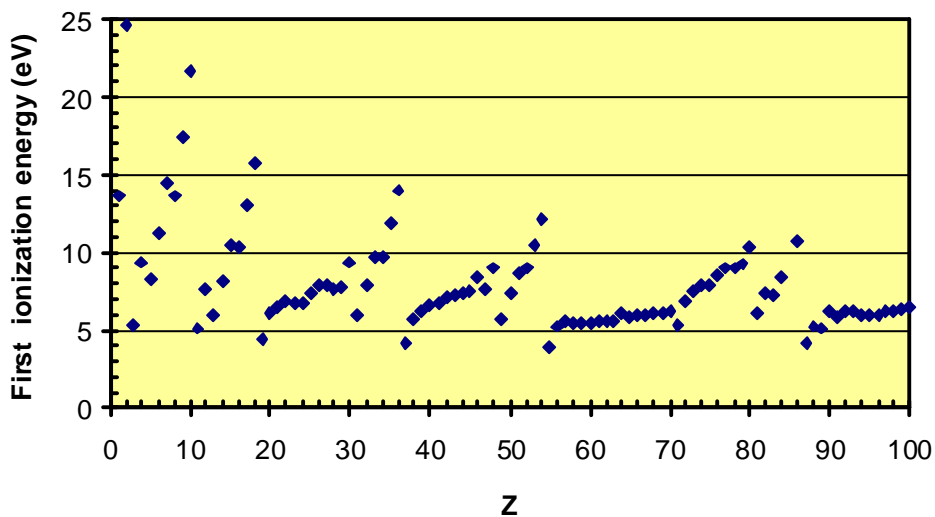
$$r = \Phi \cdot \sigma \cdot N \cdot \epsilon_{\text{target}} \cdot \epsilon_{\text{source}} \cdot \epsilon_{\text{transp}} \cdot \epsilon_{\text{det}}$$



Isotope Separation On-Line

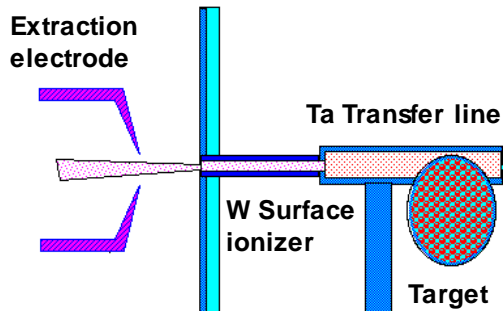
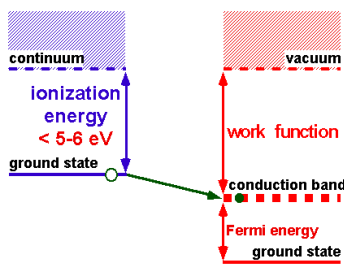
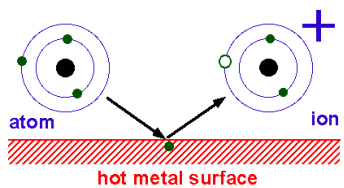


The first ionization energy of the elements



Positive surface ionization source

Surface Ionization



$$\alpha_s = g_+/g_0 \exp(-(IP - \Phi)/kT)$$

$$\epsilon_s = \alpha_s / (1 + \alpha_s)$$

Saha-Langmuir equation

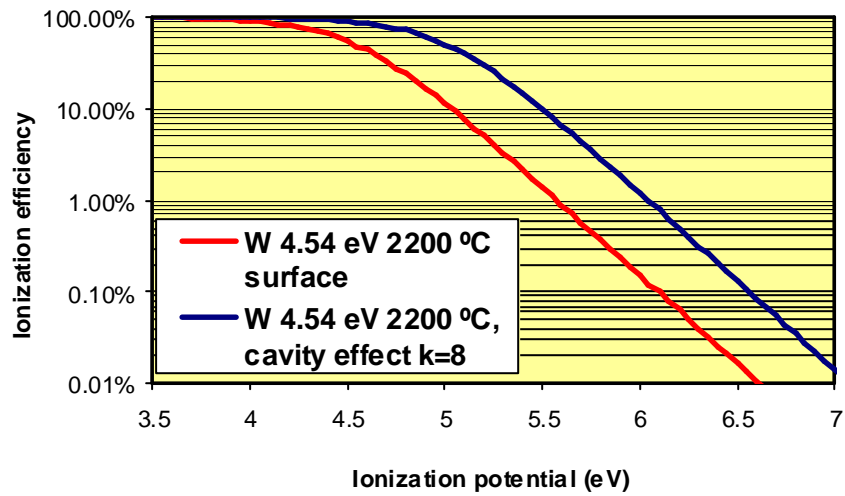
ϵ_s surface ionization efficiency

Φ work function of surface

IP ionization potential of atom

$g=2J+1$ stat. factor ($g_0=2, g_+=1$ for alkalis)

Surface ionization versus thermal ionization



$$\epsilon_{th} = 1 / (1 + g_0/g_+ / k \exp((IP - \Phi) / kT))$$

Thermal ionization efficiency in realistic ionizer cavity

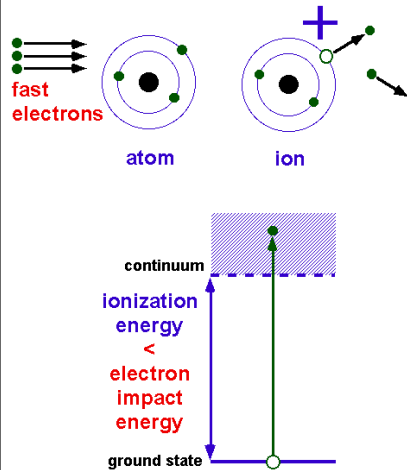
R. Kirchner, Nucl. Instr. Meth. A292 (1987) 204.

Ionization potentials of the elements

Ionization potential: < 5 eV																					
Ionization potential: 5.0 - 5.8 eV																					
Ionization potential: 5.8 - 6.5 eV																					
1																	2				
H																	He				
3	4															5	6	7	8	9	10
Li	Be															B	C	N	O	F	Ne
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19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36				
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr				
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54				
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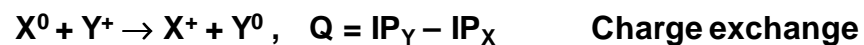
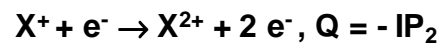
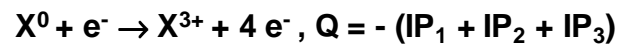
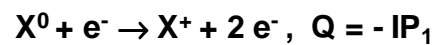
Ingredients of a plasma ion source

Ionization by electron impact

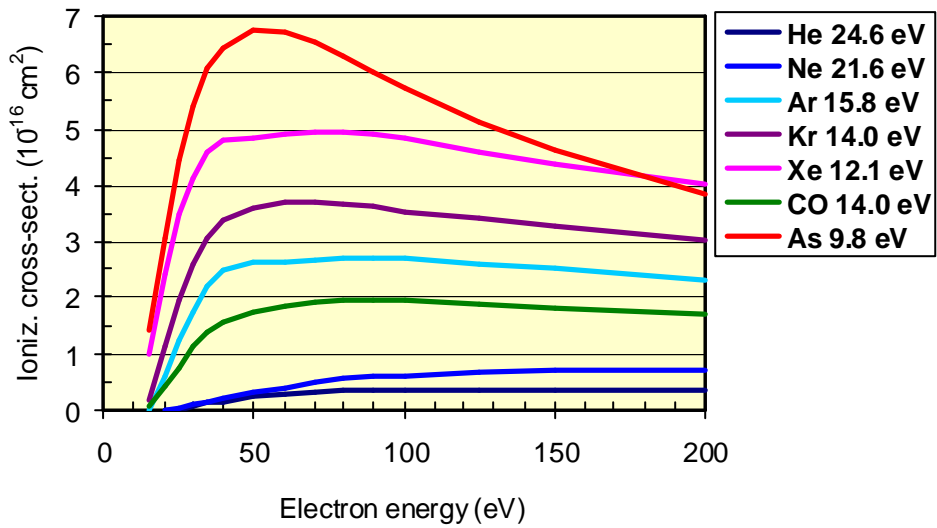


- Fast electrons:
 - A) Thermionic emission + accelerating field
 - B) RF heating
 - Atom confinement: plasma chamber
 - Electron “recycling”: magnetic field
 - Ion extraction system
- $I[\text{A}] = A^* T[\text{K}]^2 \exp(-\Phi[\text{eV}]/kT[\text{K}])$
 $A^* = 120 \text{ A cm}^{-2} \text{ K}^{-2}$
 $\nu_{\text{cyc}}[\text{GHz}] = 28 B[\text{T}]$
 $r[\text{mm}] = 0.35 E_e[\text{eV}]^{1/2}/B[\text{T}]$

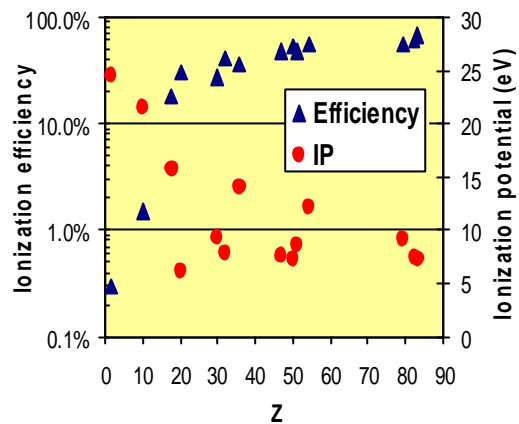
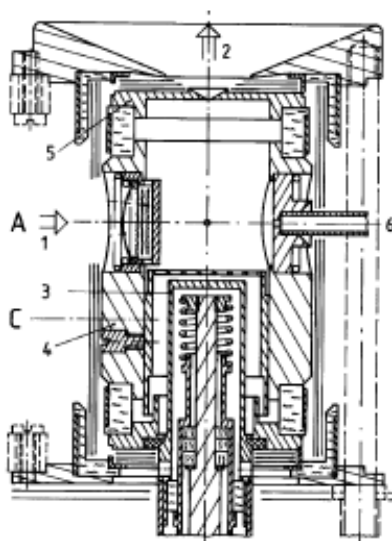
Ionization and neutralization



Electron impact ionization cross-sections



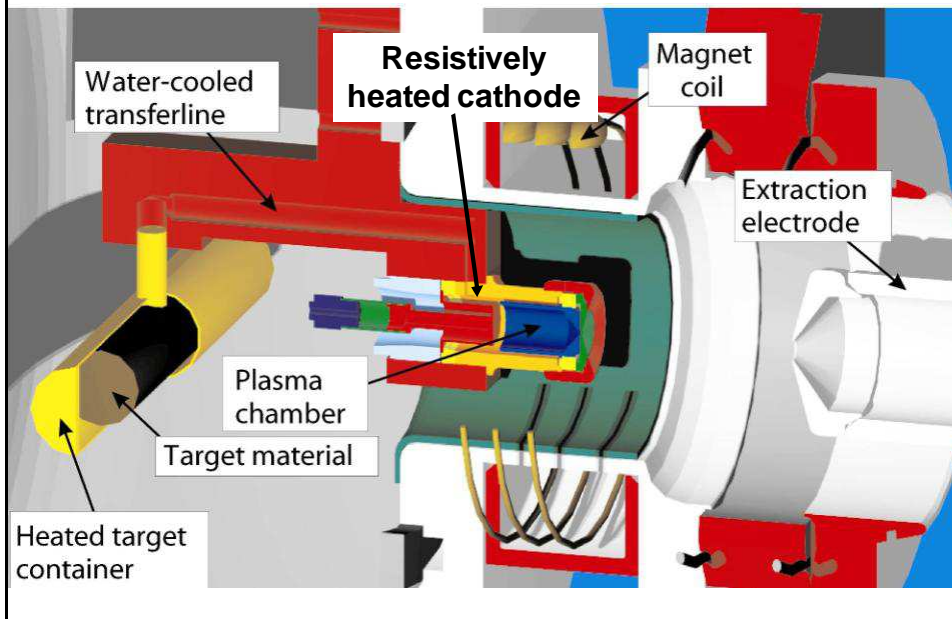
Forced Electron Beam Ion Arc Discharge (FEBIAD)



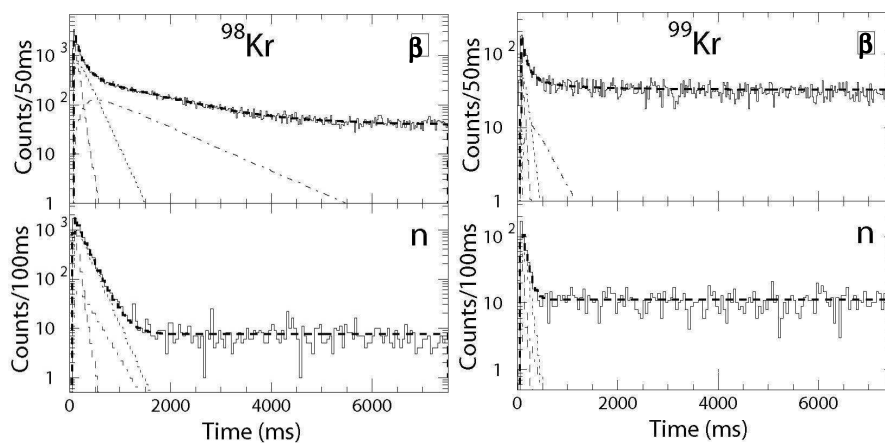
FEBIAD ion sources are excellent for heavier elements!

R. Kirchner, Rev. Sci. Instr. 67 (1996) 928.

ISOLDE "FEBIAD"



2001: $^{94-99}\text{Kr}$ decay studied at ISOLDE

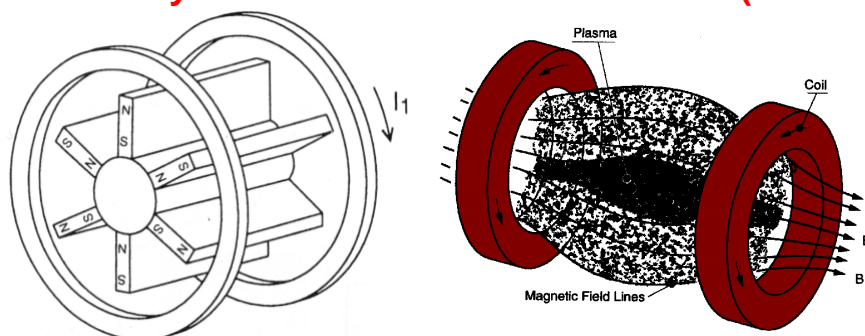


U.C. Bergmann et al., Nucl. Phys. A 714 (2003) 21.

Volatility of the elements

T (p vapor > 0.01 mbar) < 100 °C																					
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Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr								

Electron Cyclotron Resonance Ion Source (ECRIS)



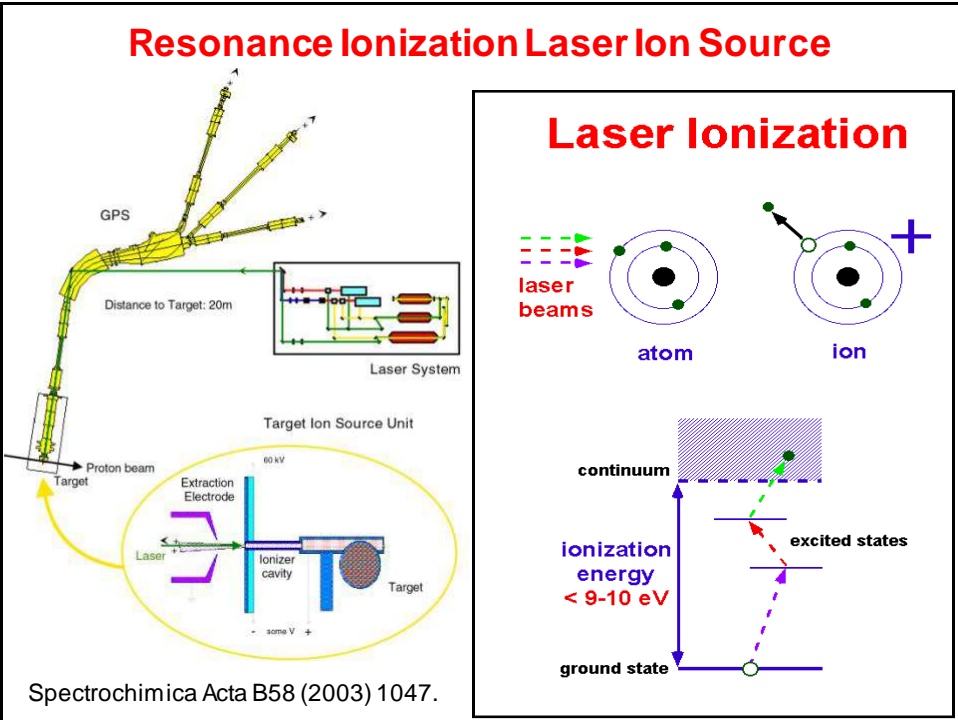
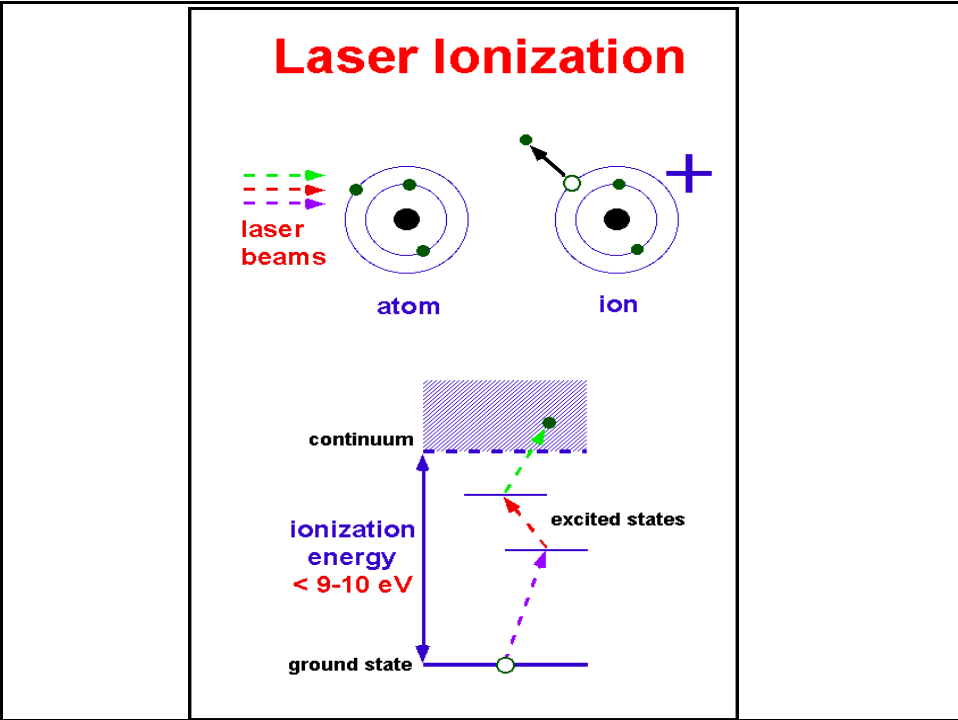
radial plasma confinement by magnetic multipole field

longitudinal plasma confinement by magnetic bottle effect (1+ ECRIS)
or minimum B configuration (n+ ECRIS)

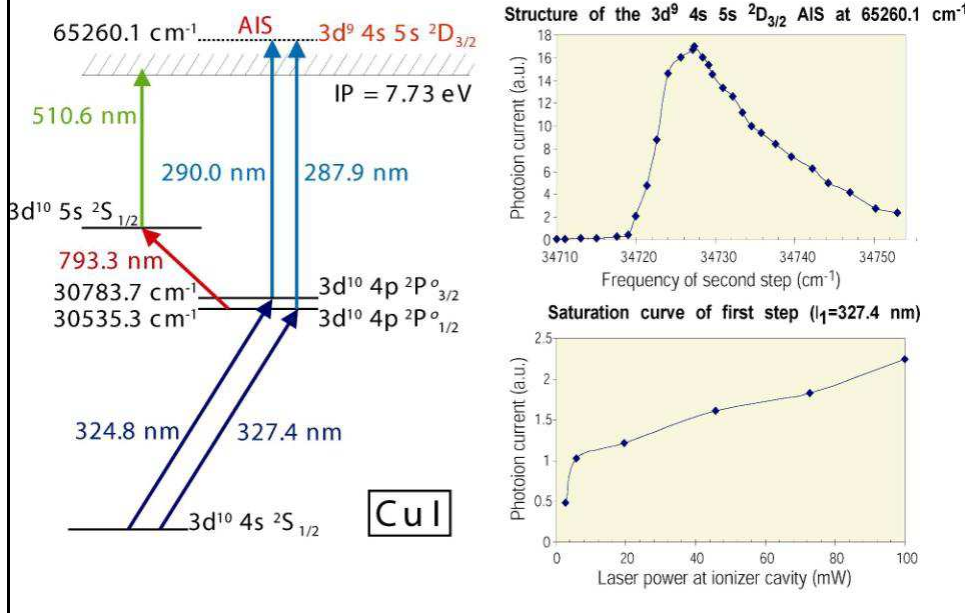
plasma heating by RF (typically 2.45 – 30 GHz)

good efficiency for light elements (20% He⁺, 50% C⁺, O⁺, Ar⁺, 90% Xe⁺)

R. Geller, Electron Cyclotron Resonance Ion Sources and ECR Plasmas, IOP, Bristol, 1996.



Ionization of Cu



Neutron-rich Mn isotopes from UC_x /graphite target

Ge 64 64 s	Ge 65 31 s	Ge 66 2.3 h	Ge 67 18.7 m	Ge 68 270.82 d	Ge 69 39.0 h	Ge 70 21.23	Ge 71 11.43 d	Ge 72 27.66	Ge 73 7.73	Ge 74 35.94	Ge 75 47 s 83 m	Ge 76 7.44
Ga 63 31.4 s	Ga 64 2.62 m	Ga 65 15 m	Ga 66 9.4 h	Ga 67 78.3 h	Ga 68 67.63 m	Ga 69 60.108	Ga 70 21.15 m	Ga 71 39.892	Ga 72 14.1 h	Ga 73 4.86 h	Ga 74 8.1 m	Ga 75 2.1 m
Zn 62 9.13 h	Zn 63 38.1 m	Zn 64 48.6	Zn 65 244.3 d	Zn 66 27.9	Zn 67 4.1	Zn 68 18.8	Zn 69 13.8 h 98 m	Zn 70 0.6	Zn 71 3.9 h 2.4 m	Zn 72 46.5 h	Zn 73 5.8 s 23.8 s	Zn 74 98 s
Cu 61 3.4 h	Cu 62 9.74 m	Cu 63 69.17	Cu 64 12.700 h	Cu 65 30.83	Cu 66 5.1 m	Cu 67 61.9 h	Cu 68 3.0 m 39 s	Cu 69 3.0 m	Cu 70 4.2 s 5 s	Cu 71 19.5 s	Cu 72 6.6 s	Cu 73 3.9 s
Ni 60 26.223	Ni 61 1.140	Ni 62 3.634	Ni 63 100 a	Ni 64 0.926	Ni 65 2.52 h	Ni 66 54.6 h	Ni 67 21 s	Ni 68 29 s	Ni 69 11.4 s	Ni 70 6.0 s	Ni 71 2.56 s	Ni 72 1.57 s
Co 59 100	Co 60 16.5 m 8.27 s	Co 61 1.65 h	Co 62 14.9 m 1.8 m	Co 63 27.5 s	Co 64 0.3 s	Co 65 1.14 s	Co 66 0.23 s	Co 67 0.42 s	Co 68 0.18 s	Co 69 0.27 s	Co 70 0.15 s	Co 71 0.21 s
Fe 58 0.28	Fe 59 44.503 d	Fe 60 $1.5 \cdot 10^6$ a	Fe 61 6.0 m	Fe 62 68 s	Fe 63 6.1 s	Fe 64 2.0 s	Fe 65 0.45 s	Fe 66 0.44 s	Fe 67 0.47 s	Fe 68 0.1 s	Fe 69 0.17 s	44
Mn 57 1.5 m	Mn 58 98 s 2.8 s	Mn 59 4.6 s	Mn 60 1.77 s 91 s	Mn 61 623 ms	Mn 62 671 ms	Mn 63 275 ms	Mn 64 89 ms	Mn 65 88 ms	Mn 66 66 ms	Mn 67 42 ms	Mn 68 28 ms	Mn 69 14 ms

M. Hannawald et al., Phys. Rev. Lett. 82 (1999) 1391.

Surface ionized background

1	Ionization potential: < 5 eV																2	
3	Ionization potential: 5.0 - 5.8 eV																	
11	Ionization potential: 5.8 - 6.5 eV																	
19																		
37																		
55																		
87																		

58	59	60	61	62	63	64	65	66	67	68	69	70	71
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu
90	91	92	93	94	95	96	97	98	99	100	101	102	103
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr

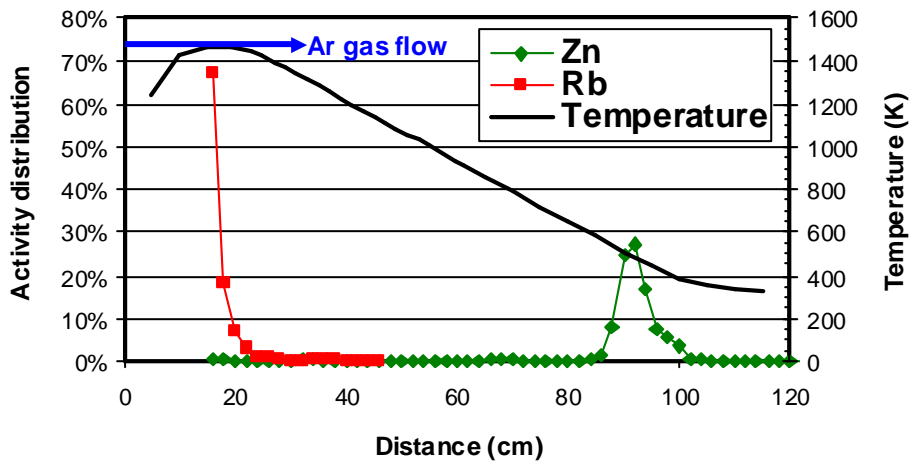
ISOLDE beams around N=50

Sr 76 2.69 m	Sr 78 2.3 m	Sr 80 1.8 h	Sr 81 22.2 m	Sr 82 25.34 d	Sr 83 5.3 s	Sr 84 627 m	Sr 85 643 d	Sr 86 9.88 s	Sr 87 2.01 s	Sr 88 82.58 s	Sr 89 60.5 d	Sr 90 28.64 a	Sr 91 9.5 h	Sr 92 2.71 h
Rb 77 3.9 m	Rb 78 30.0 m	Rb 79 30 s	Rb 80 30 s	Rb 81 4.98 h	Rb 82 1.27 h	Rb 83 80.2 d	Rb 84 20.5 m	Rb 85 72.165 s	Rb 86 1.02 m	Rb 87 24.335 s	Rb 88 17.3 m	Rb 89 15.2 m	Rb 90 4.8 m	Rb 91 58 s
Kr 76 14.6 h	Kr 77 1.24 h	Kr 78 0.35 s	Kr 79 59 s	Kr 80 2.25 s	Kr 81 33 s	Kr 82 11.8 s	Kr 83 11.5 s	Kr 84 57.0 s	Kr 85 4.48 s	Kr 86 17.3 s	Kr 87 76.3 m	Kr 88 2.84 h	Kr 89 3.18 m	Kr 90 32.3 s
Br 75 1.6 h	Br 76 1.28 s	Br 77 43 m	Br 78 4.45 m	Br 79 4.81 s	Br 80 1.03 m	Br 81 45.31 s	Br 82 43 m	Br 83 2.40 h	Br 84 43 m	Br 85 2.87 m	Br 86 55.1 s	Br 87 50.7 s	Br 88 16.3 s	Br 89 4.40 s

⁸¹Rb background is 15000 times more abundant than ⁸¹Zn!

As 73 80.3 d	As 74 17.77 d	As 75 100 s	As 76 26.4 h	As 77 38.9 h	As 78 1.57 h	As 79 8.2 m	As 80 15.2 s	As 81 3.6 s	As 82 447 s	As 83 13.3 s	As 84 4.5 s	As 85 2.03 s	As 86 0.9 s	As 87 0.79 s
Ge 72 27.66 s	Ge 73 7.73 s	Ge 74 35.94 s	Ge 75 4.85 h	Ge 76 7.44 s	Ge 77 1.53 s	Ge 78 88 m	Ge 79 29 s	Ge 80 29.5 s	Ge 81 78 s	Ge 82 4.60 s	Ge 83 1.85 s	Ge 84 984 ms	Ge 85 535 ms	Ge 86 54 ms
Ga 71 39.932 s	Ga 72 14.1 h	Ga 73 4.85 h	Ga 74 1.4 h	Ga 75 2.1 m	Ga 76 32.6 s	Ga 77 13 s	Ga 78 5.49 s	Ga 79 2.85 s	Ga 80 1.70 s	Ga 81 1.22 s	Ga 82 0.60 s	Ga 83 0.31 s	Ga 84 85 ms	Ga 85 1.327 s
Zn 70 0.6 s	Zn 71 2.8 s	Zn 72 46.5 h	Zn 73 2.8 s	Zn 74 96 s	Zn 75 10.2 s	Zn 76 5.6 s	Zn 77 1.47 s	Zn 78 1.47 s	Zn 79 595 ms	Zn 80 537 ms	Zn 81 0.29 s	Zn 82 0.3239 s	Zn 83 0.5490 s	Zn 84 1.005 s
Cu 69 3.0 m	Cu 70 4.4 s	Cu 71 19.5 s	Cu 72 6.6 s	Cu 73 3.9 s	Cu 74 1.59 s	Cu 75 1.22 s	Cu 76 1.22 s	Cu 77 469 ms	Cu 78 342 ms	Cu 79 186 ms	Cu 80 1.974 s	Cu 81 0.1269 s	Cu 82 0.04712 s	Cu 83 0.1269 s
Ni 68 29 s	Ni 69 11.4 s	Ni 70 5.0 s	Ni 71 2.56 s	Ni 72 1.57 s	Ni 73 0.84 s	Ni 74 0.9 s	Ni 75 0.6 s	Ni 76 -0.24 s	Ni 77 0 s	Ni 78 0 s	Ni 79 0 s	Ni 80 0 s	Ni 81 0 s	Ni 82 0 s

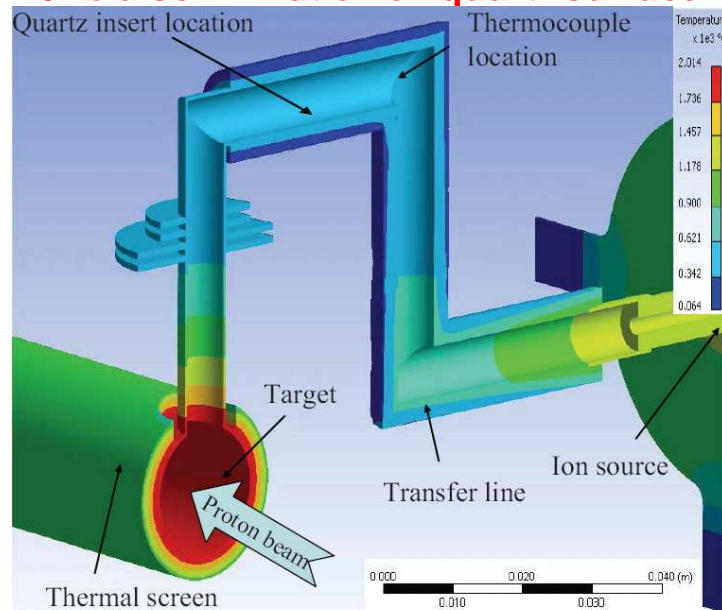
Zn/Rb discrimination on quartz surface!



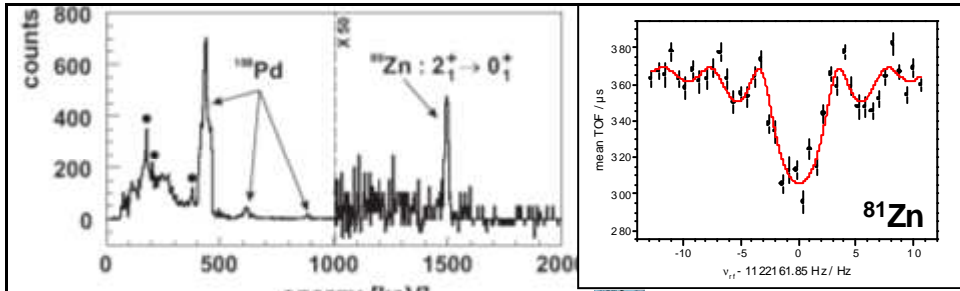
Combination of neutron converter and quartz transfer line provides $^{81}\text{Zn}/^{81}\text{Rb}$ selectivity gain of 100000!

Nucl. Instr. Meth. B266 (2008) 4229.

Zn/Rb discrimination on quartz surface!

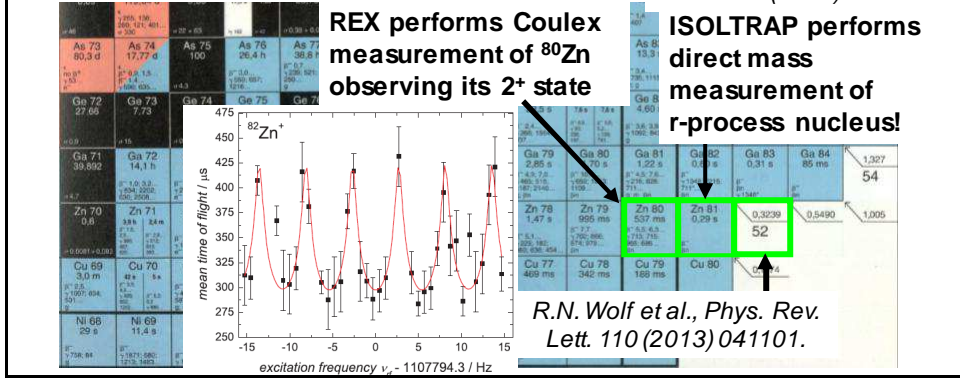


E. Boucquerel et al., Nucl. Instr. Meth. B266 (2008) 4298.

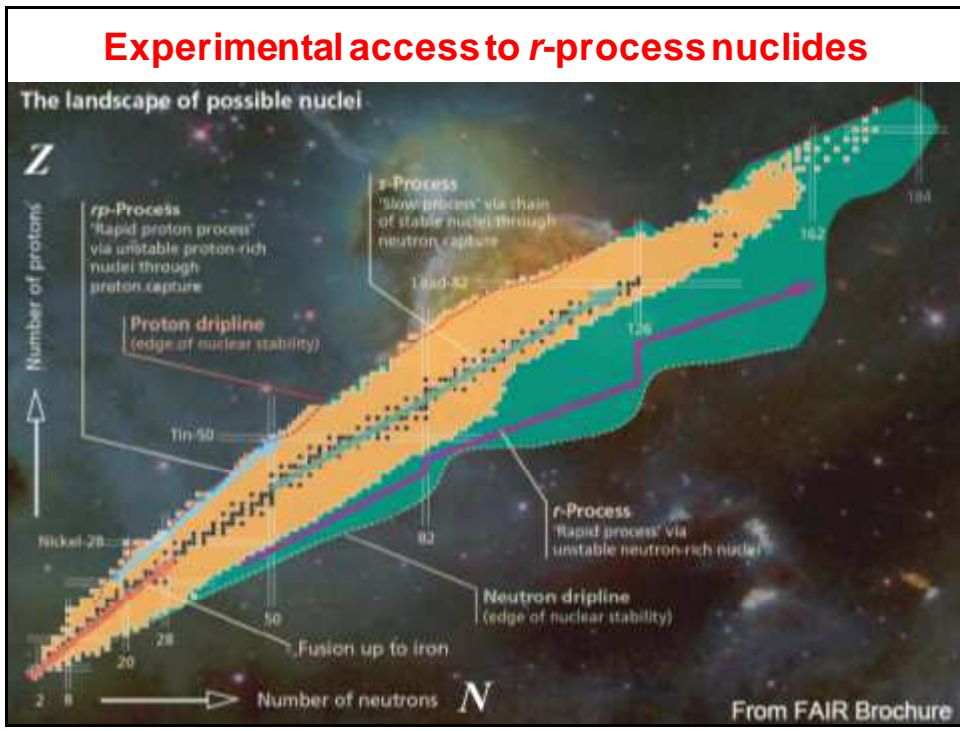


J. Van de Walle et al., Phys. Rev. Lett. 99 (2007) 142501.

S. Baruah et al., Phys. Rev. Lett. 101 (2008) 262501.



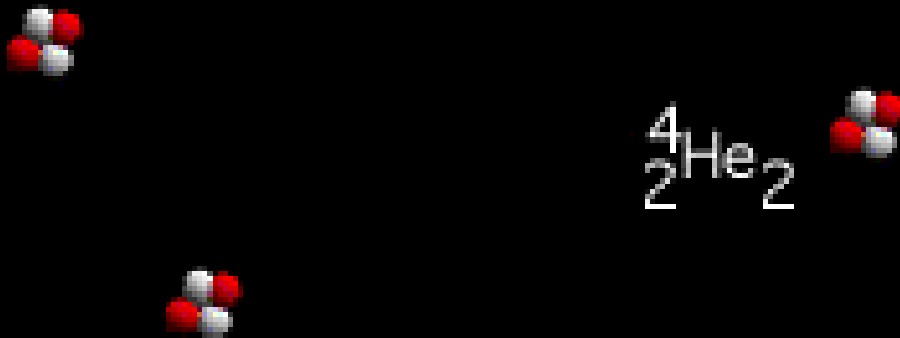
Experimental access to r-process nuclides



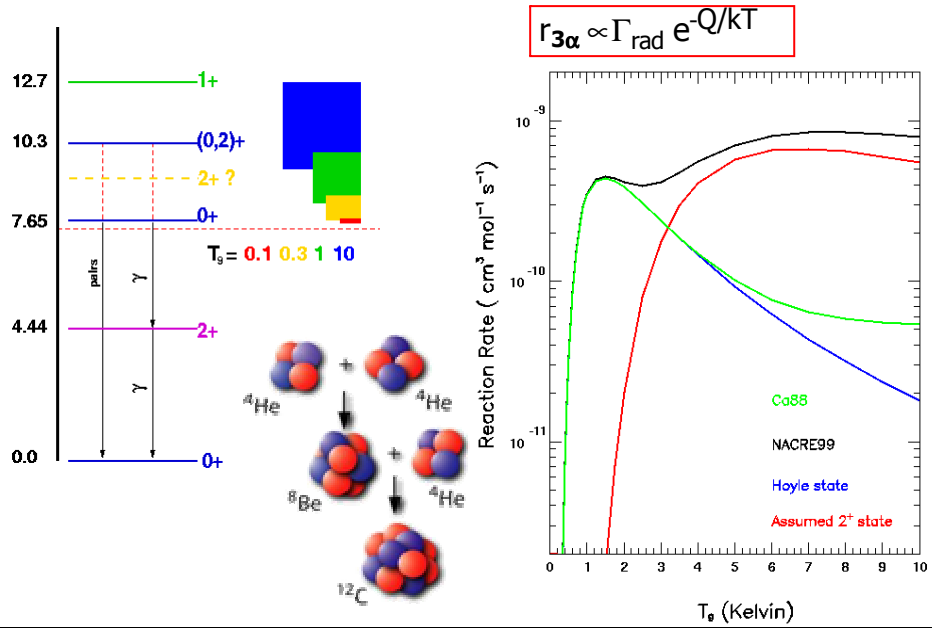
Production of ^{12}C in stars

C 8	C 9	C 10	C 11	C 12	C 13	C 14
2E-21 s	127 ms	19.3 s	20 m			5.7 ka
B 7	B 8	B 9	B 10	B 11	B 12	B 13
4E-24 s	770 ms	8E-19 s			20 ms	17 ms
Be 6	Be 7	Be 8	Be 9	Be 10	Be 11	Be 12
5E-21 s	53.3 d	7E-17 s		1.5 Ma	13.8 s	21 ms
Li 5	Li 6	Li 7	Li 8	Li 9	Li 10	Li 11
4E-22 s			840 ms	178 ms	2E-21 s	8.5 ms
He 3	He 4	He 5	He 6	He 7	He 8	He 9
		7E-22 s	807 ms	3E-21 s	119 ms	7E-21 s
H 1	H 2	H 3				
		12.3 a				

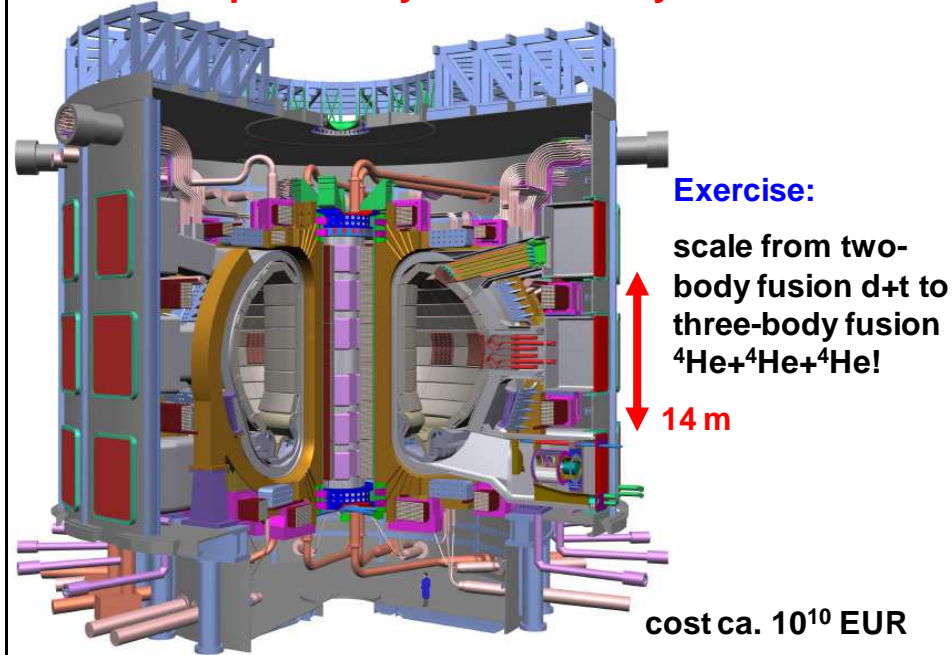
The triple-alpha process: rate



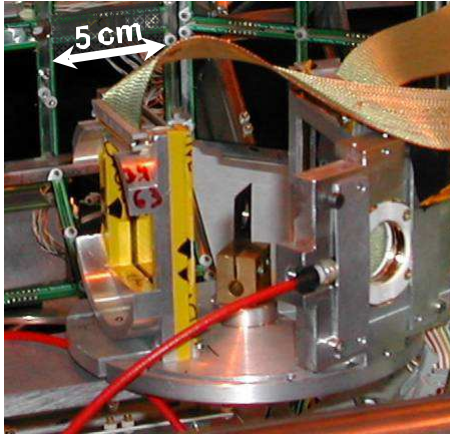
The triple-alpha process



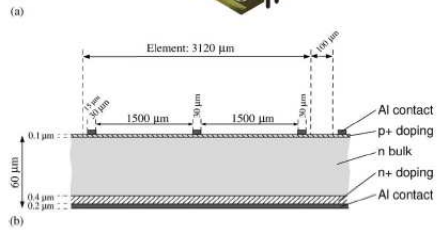
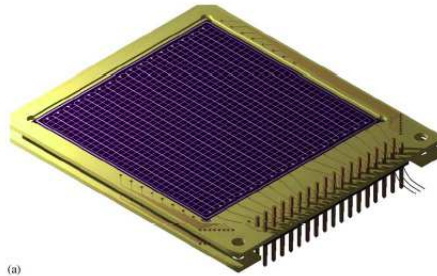
Setup for study of three-body-fusion?



Setup for study of triple alpha reaction!



New detector design

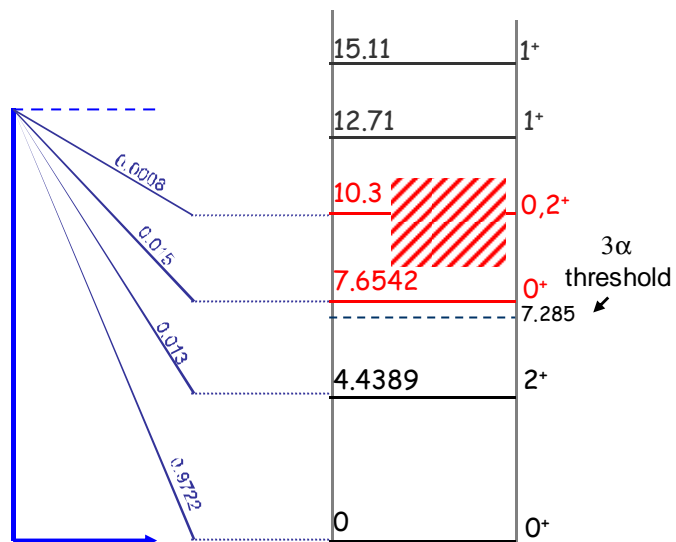


Reduced deadlayer

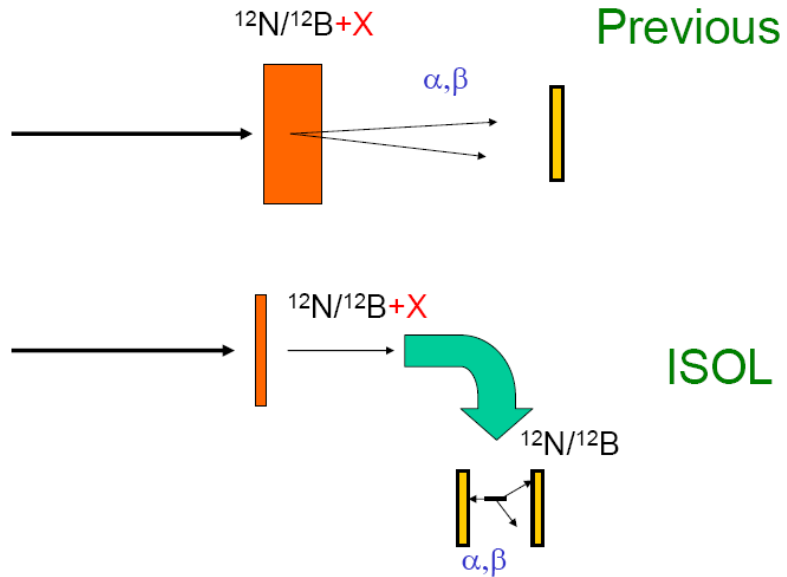
Inverse reaction: $^{12}\text{B}(\beta, 3\alpha)$ decay

^{12}B 1+

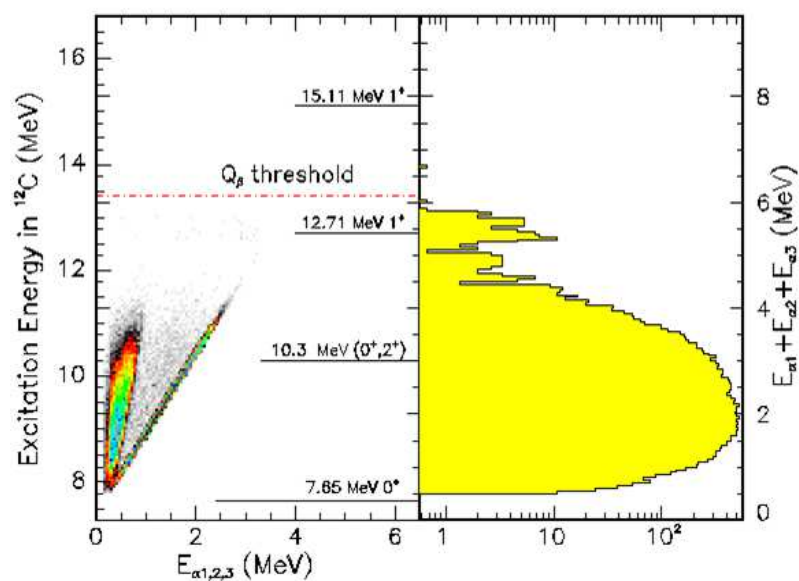
B 12
20.2 ms



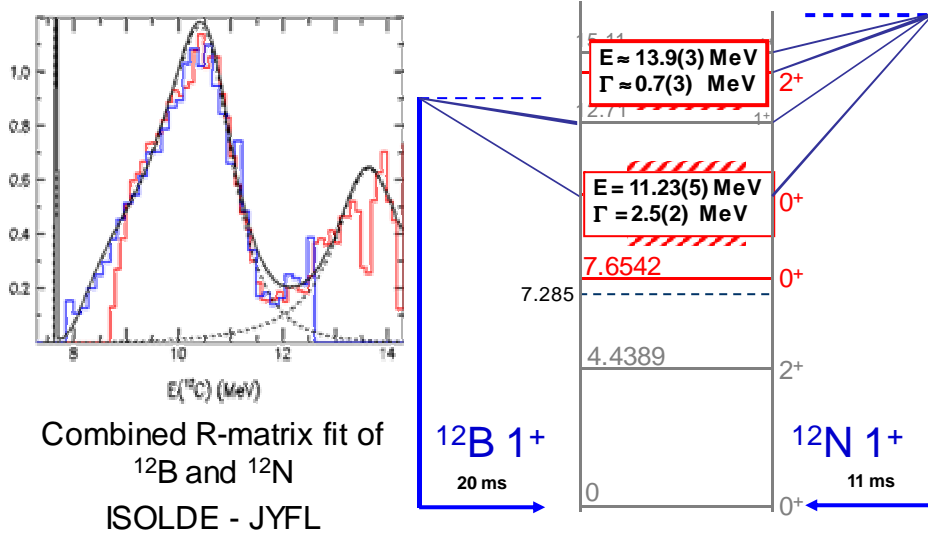
How to measure beta-delayed particle emission?



$^{12}\text{Be}(\beta^-)^{12}\text{B}$ beta decay to $^{12}\text{C}^* \rightarrow 2\alpha$ detected

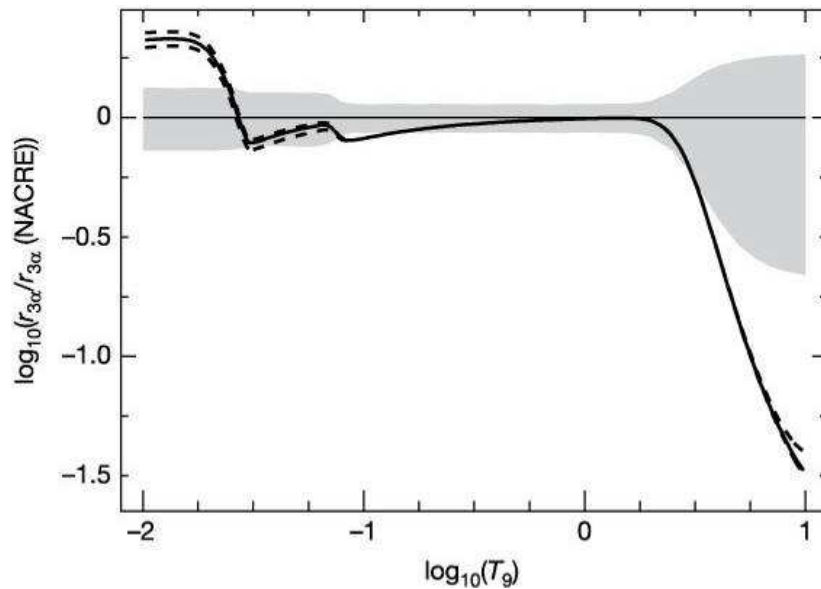


The triple-alpha process: ^{12}B and ^{12}N decays



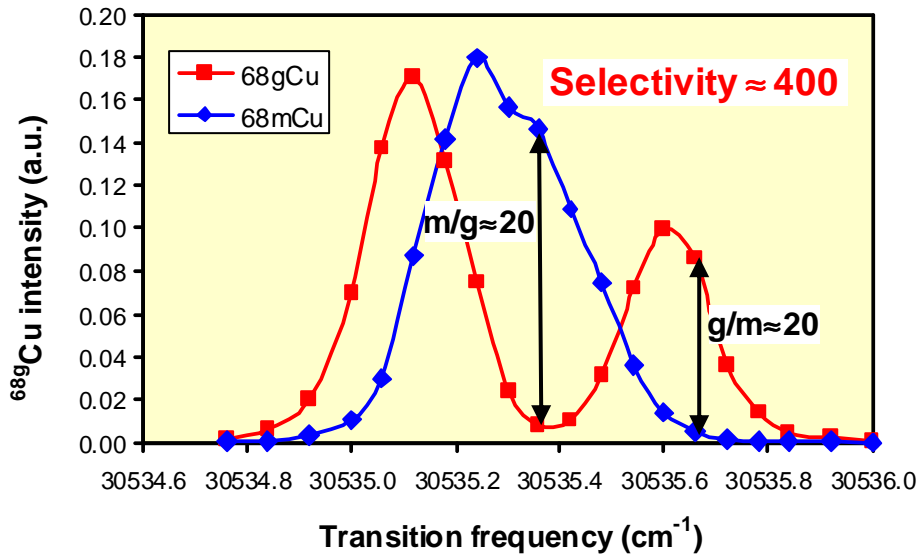
H.O.U. Fynbo et al., Nature 433 (2005) 136.

New rates for the triple-alpha process



H.O.U. Fynbo et al., Nature 433 (2005) 136.

Isomer separation



Mass measurements with Penning traps

\vec{B}

q/m

Cyclotron frequency: $f_c = \frac{1}{2\pi} \cdot \frac{q}{m} \cdot B$

PENNING trap

- Strong homogeneous magnetic field
- Weak electric 3D quadrupole field

end cap

ring electrode

V

z_c

B

$\omega_+ + \omega_- = \omega_c = \frac{q}{m} B$

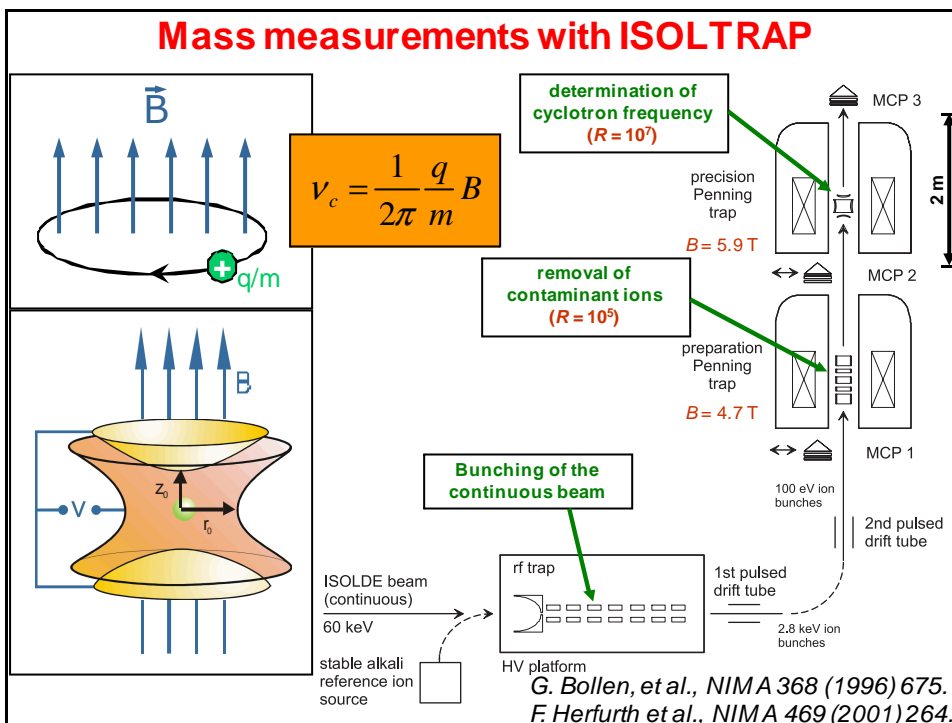
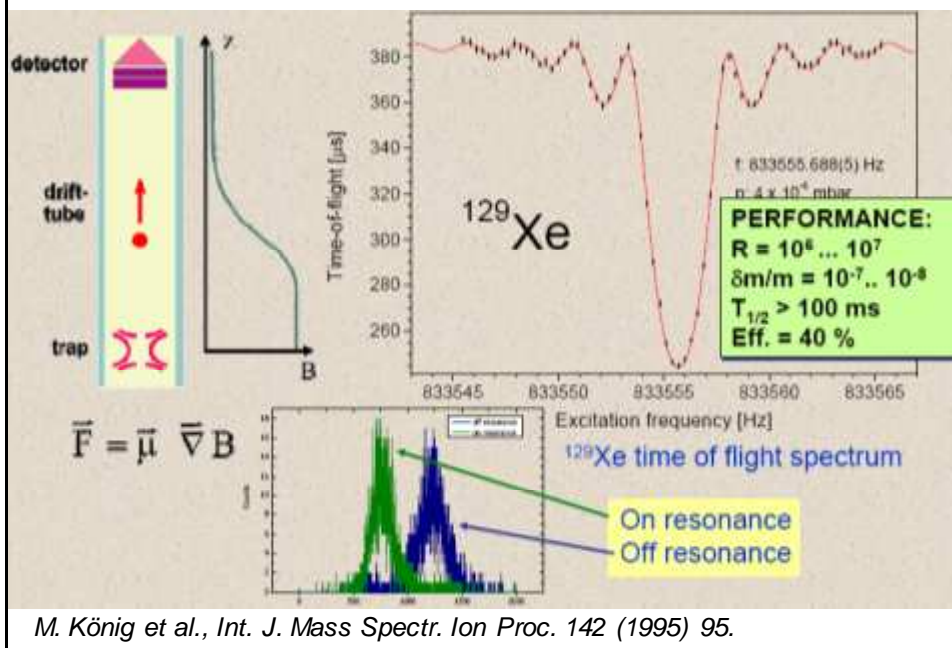
axial (z)

magnetron (-)

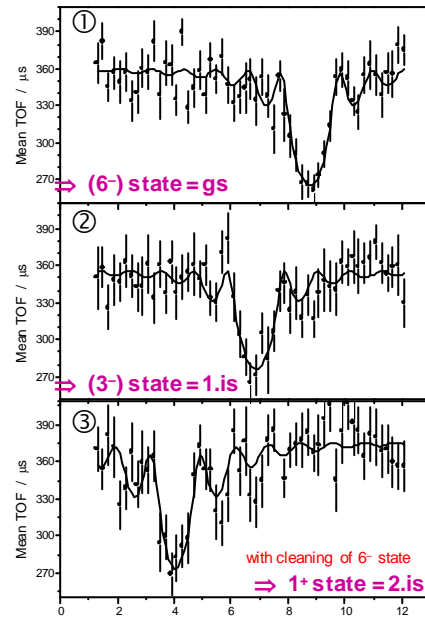
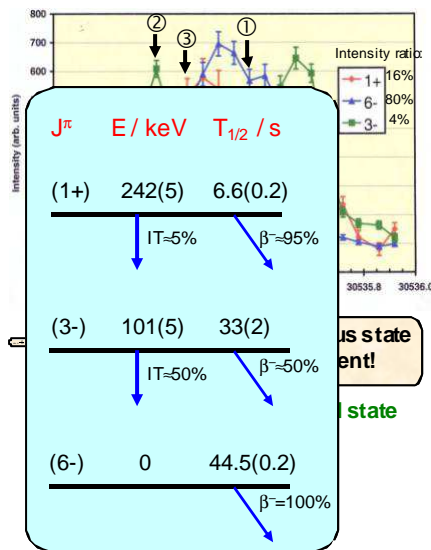
cyclotron (+) radial

Nobel prize 1989:
Dehmelt

Resonance frequency measurement via TOF method

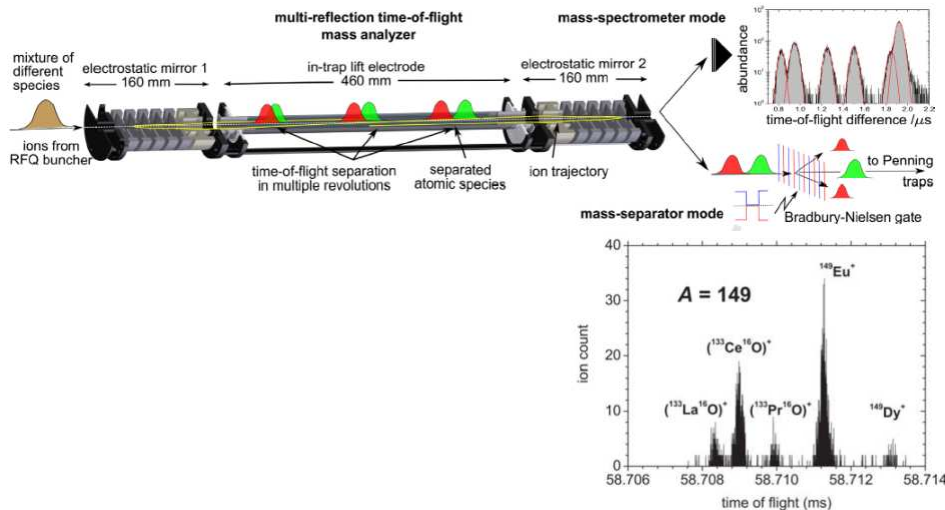


Solving the ^{70}Cu mass puzzle



J. Van Roosbroeck et al., Phys. Rev. Lett. 92 (2004) 112501.

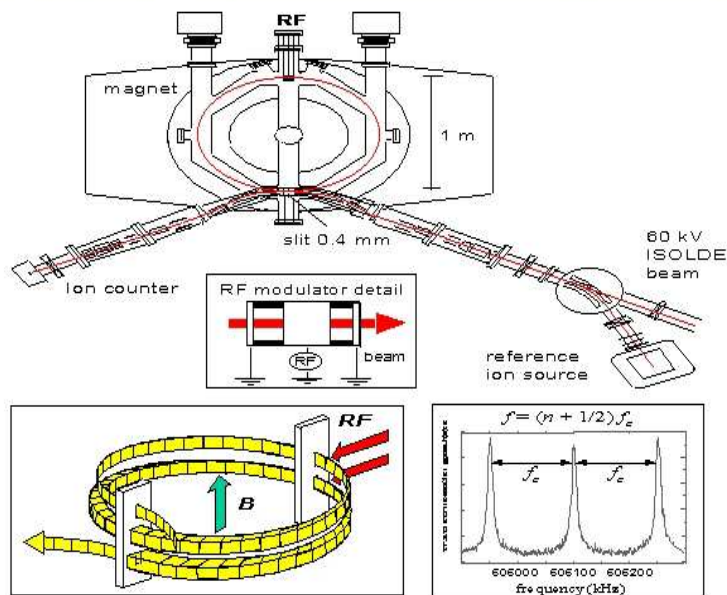
Linear TOF spectrometer



R.N. Wolf et al., Nucl. Instr. Meth. A686 (2012) 82.

R.N. Wolf et al., Int. J. Mass Spectrometry (2013) in press.

S. Kreim et al., Nucl. Instr. Meth. B, Proc. of EMIS-16, in press.



M. de Saint Simon et al., *Phys. Scr.* T59 (1995) 406.

Mass measurement of ^{11}Li ($T_{1/2} = 9$ ms)

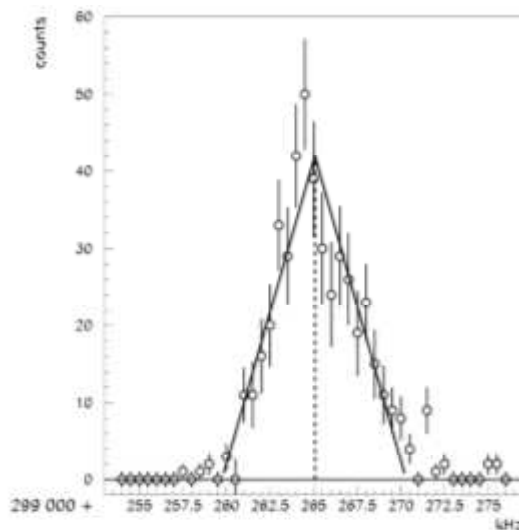
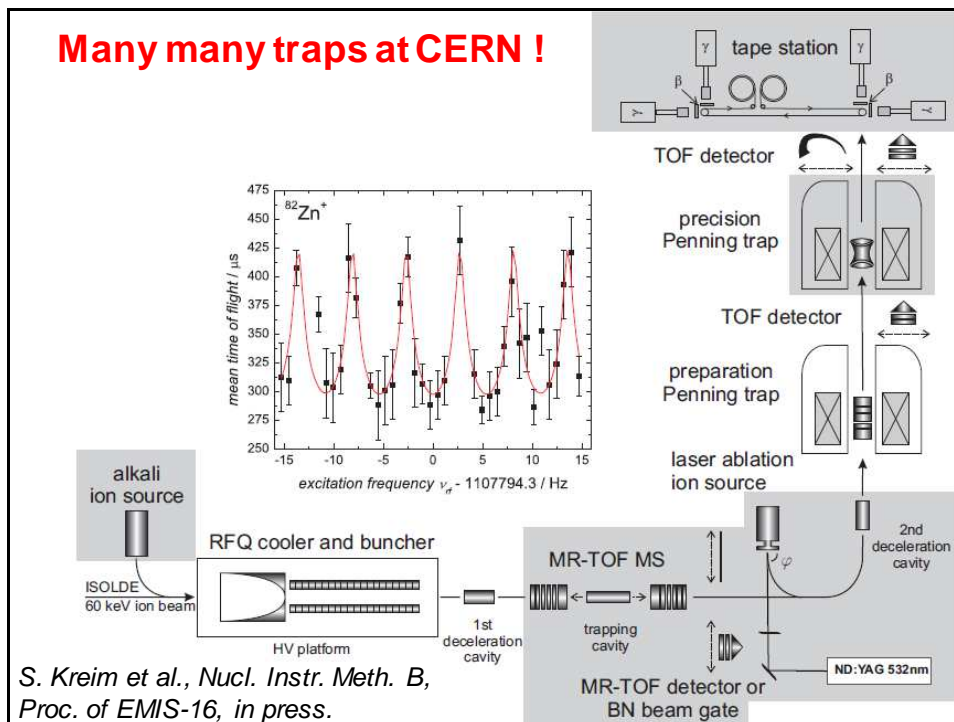


Fig. 2. Example of accumulated transmission peak for ^{11}Li . The center position (299 265 kHz) corresponds to the 917th harmonic of the cyclotron frequency. With a RF power of 100 W, a mass resolving power of $\frac{\Delta M}{M} \sim 57000$ was achieved.

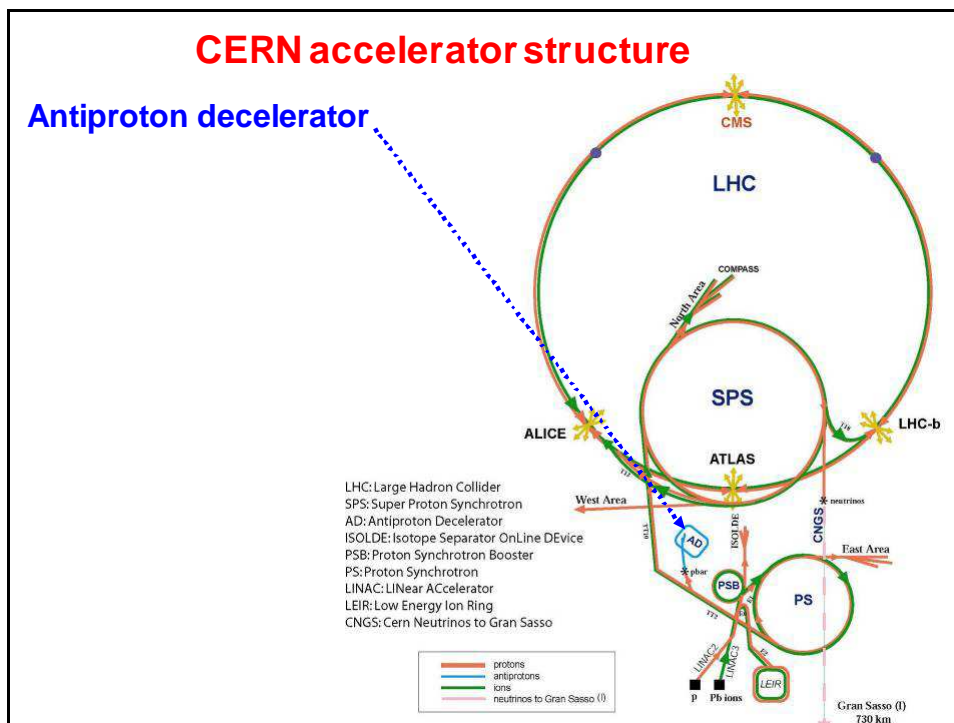
C. Bachelet et al., *Phys. Rev. Lett.* 100 (2008) 182501.

Many many traps at CERN !

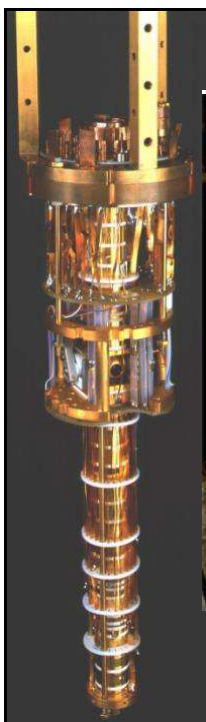


CERN accelerator structure

Antiproton decelerator



Antiproton traps at CERN



G. Gabrielse et al., *Phys. Rev. Lett.* 100 (2008) 113001.

M. Amoretti et al., *Phys. Lett. B* 583 (2004) 59.

D. Brown, R. Howard et al., "Angels and Demons" (2009).

Elements ionizable with CVL or Nd-YAG pumped dye or Ti:Sa lasers

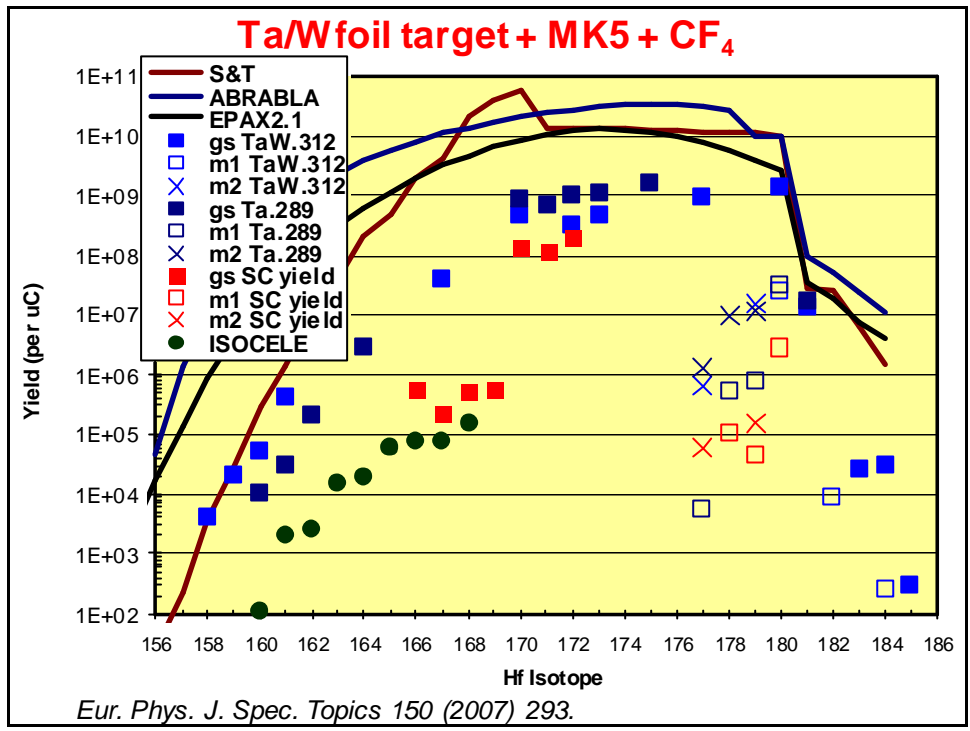
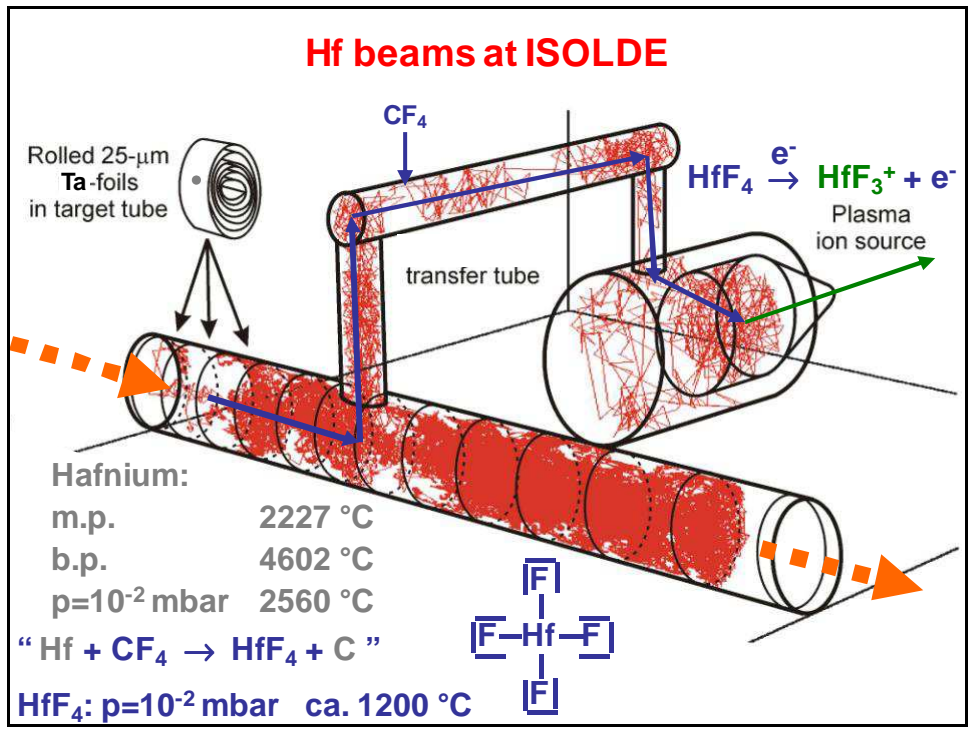
elements ionized with ISOLDE RILIS

tested ionization scheme

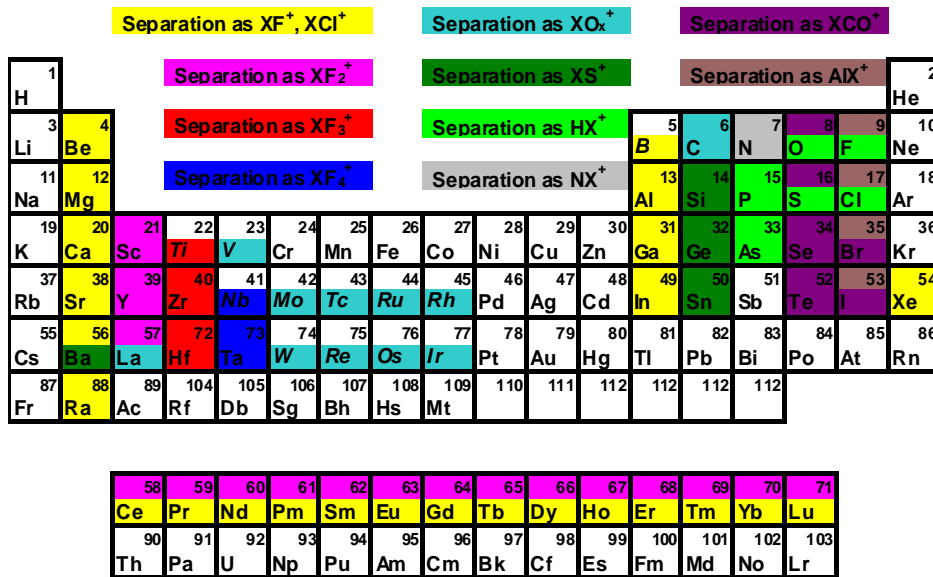
possible ionization scheme (untested)

refractory elements

1																	2
H																	He
3	4															10	
Li	Be															Ne	
11	12															18	
Na	Mg															Ar	
19	20	21	22	23	24	25	26	27	28	29	30	31	32	33	34	35	36
K	Ca	Sc	Ti	V	Cr	Mn	Fe	Co	Ni	Cu	Zn	Ga	Ge	As	Se	Br	Kr
37	38	39	40	41	42	43	44	45	46	47	48	49	50	51	52	53	54
Rb	Sr	Y	Zr	Nb	Mo	Tc	Ru	Rh	Pd	Ag	Cd	In	Sn	Sb	Te	I	Xe
55	56	57	72	73	74	75	76	77	78	79	80	81	82	83	84	85	86
Cs	Ba	La	Hf	Ta	W	Re	Os	Ir	Pt	Au	Hg	Tl	Pb	Bi	Po	At	Rn
87	88	89	104	105	106	107	108	109	110	111	112						
Fr	Ra	Ac	Rf	Db	Sg	Bh	Hs	Mt									
58	59	60	61	62	63	64	65	66	67	68	69	70	71				
Ce	Pr	Nd	Pm	Sm	Eu	Gd	Tb	Dy	Ho	Er	Tm	Yb	Lu				
90	91	92	93	94	95	96	97	98	99	100	101	102	103				
Th	Pa	U	Np	Pu	Am	Cm	Bk	Cf	Es	Fm	Md	No	Lr				



Overview of molecular ISOL beams



Nucl. Instr. Meth. B266 (2008) 4229.

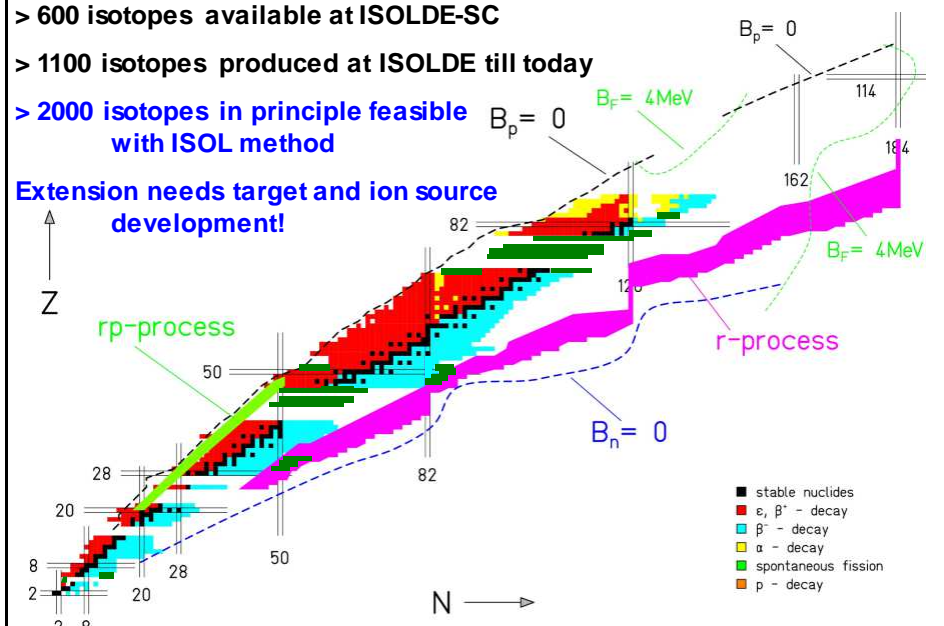
Nuclear chart at ISOLDE

> 600 isotopes available at ISOLDE-SC

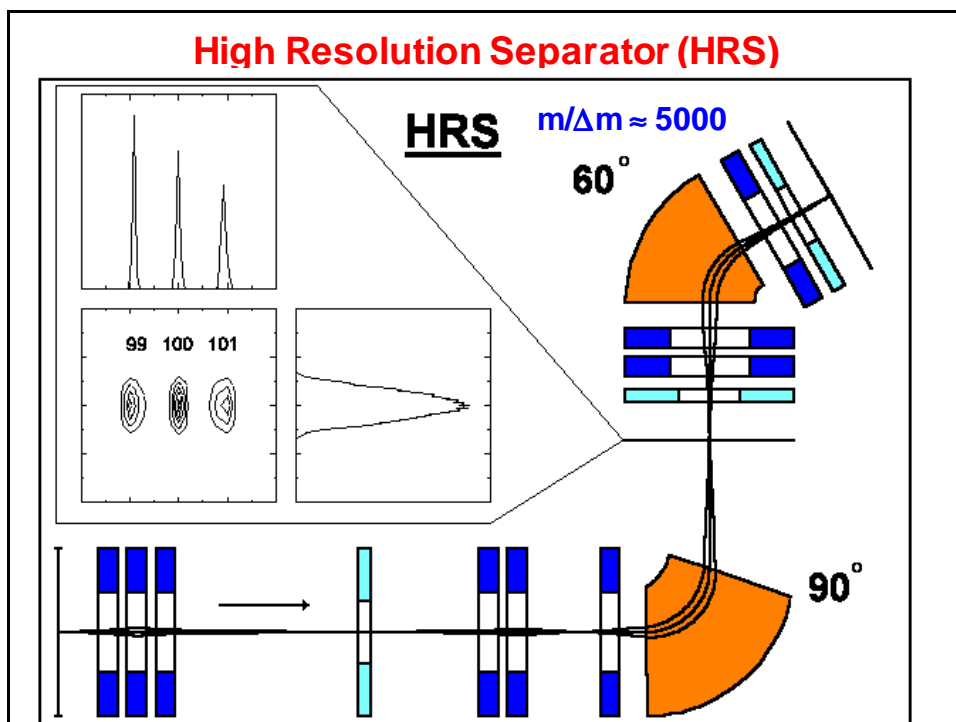
> 1100 isotopes produced at ISOLDE till today

> 2000 isotopes in principle feasible with ISOL method

Extension needs target and ion source development!

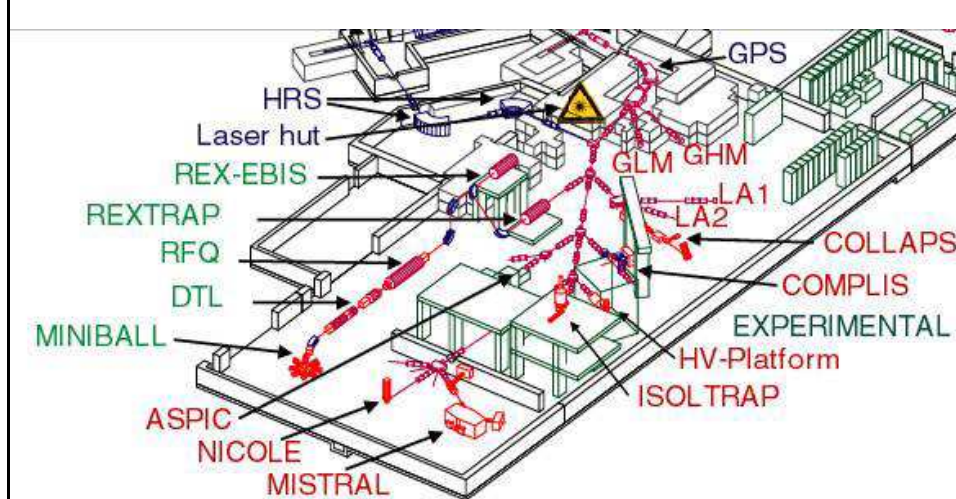


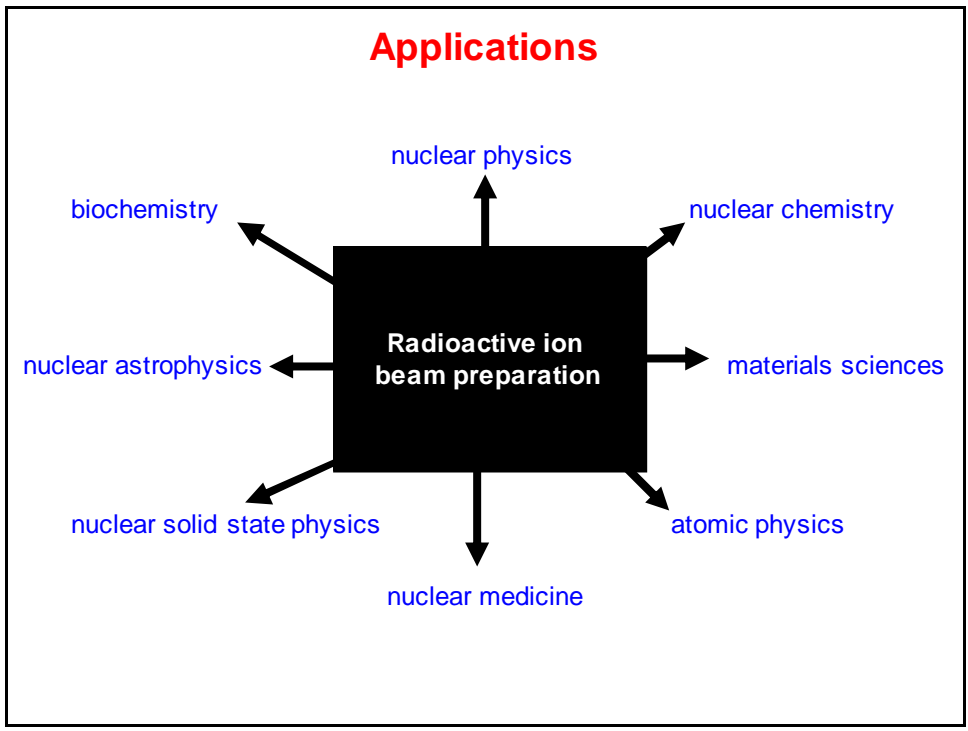
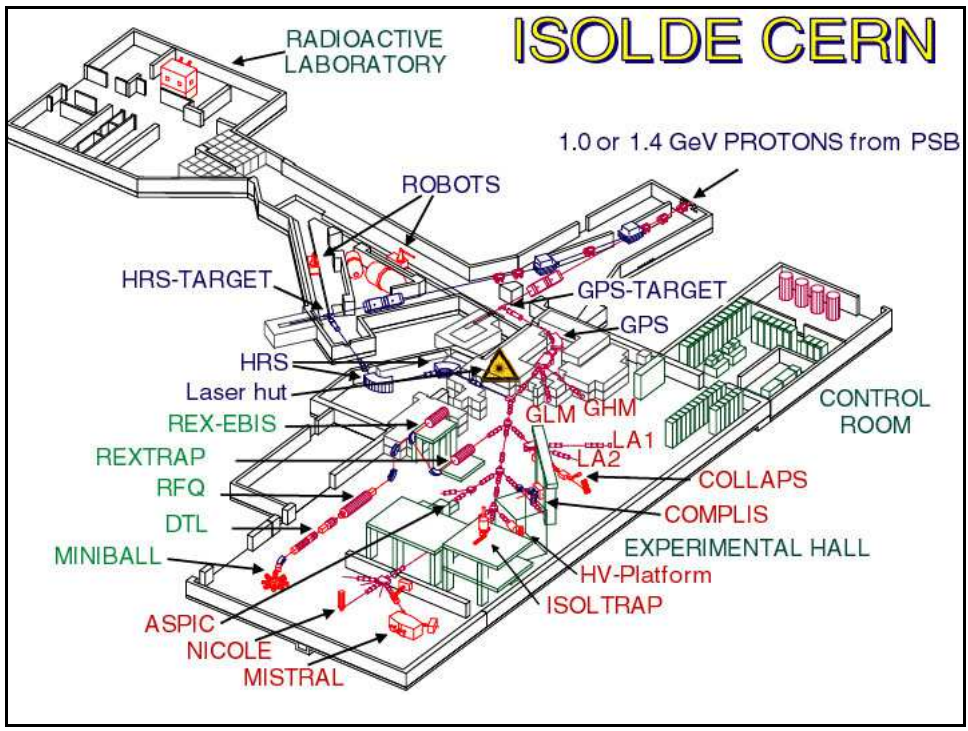
High Resolution Separator (HRS)



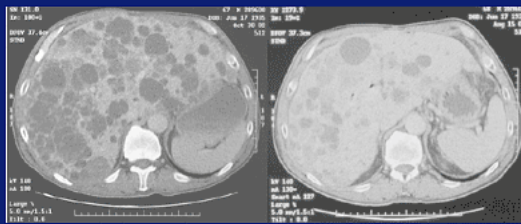
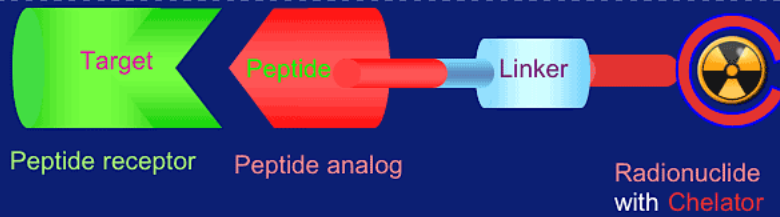
Beam transport

electrostatic beam transport is mass-independent ($E=60$ keV),
 but has space charge limit for high beam intensities ($>10 \mu\text{A}$)
 \Rightarrow high current beams need magnetic beam transport





Somatostatin analogues: Peptide Receptor Radionuclide Therapy (PRRT)



[¹⁷⁷Lu-DOTA, Tyr³]octreotate

Roelf Valkema, EANM-2008.



Radionuclides for RIT and PRRT

Radio-nuclide	Half-life	E mean (keV)	E _γ (B.R.) (keV)	Range
Y-90	64 h	934 β	-	12 mm
I-131	8 days	182 β	364 (82%)	3 mm
Lu-177	7 days	134 β	208 (10%) 113 (6%)	2 mm
Tb-161	7 days	154 β 5, 17, 40 e ⁻	75 (10%)	2 mm 1-30 μm
Tb-149	4.1 h	3967 α	165,...	25 μm
Ge-71	11 days	8 e ⁻	-	1.7 μm
Er-165	10.3 h	5.3 e ⁻	-	0.6 μm

cross-fire

Established isotopes

Emerging isotopes

R&D isotopes: supply-limited!

localized

Modern, better targeted bioconjugates require shorter-range radiation ⇒ need for **adequate (R&D) radioisotope supply**.

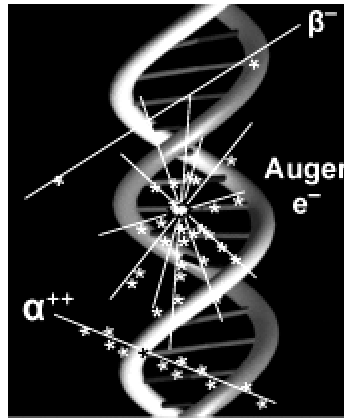
Isotopes for targeted alpha therapy



^{149}Tb for targeted alpha therapy

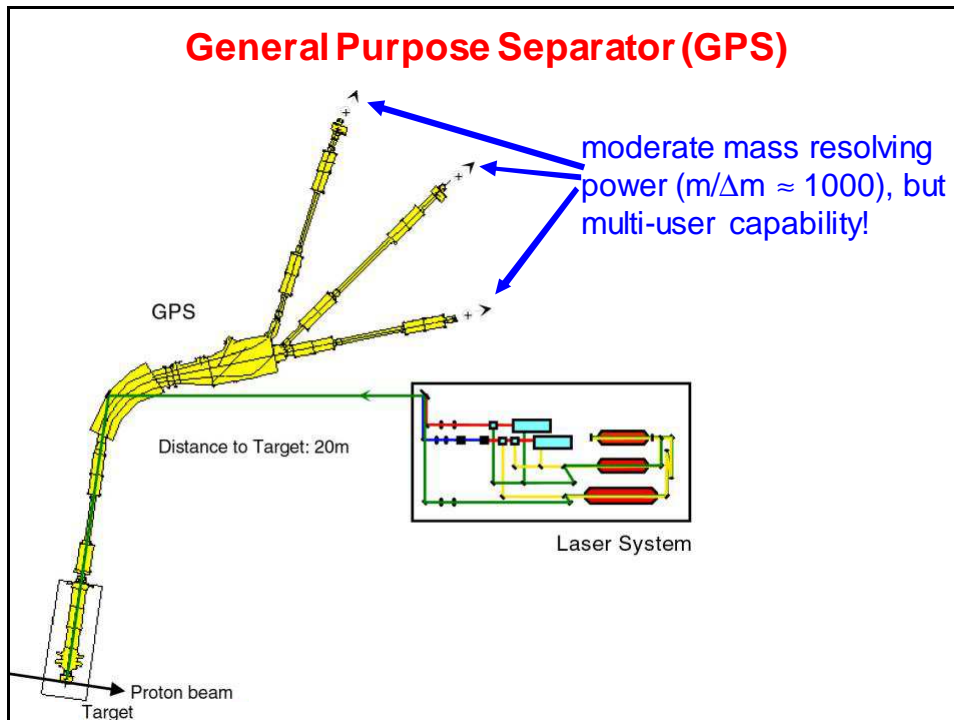


Terbium: a unique element for nuclear medicine

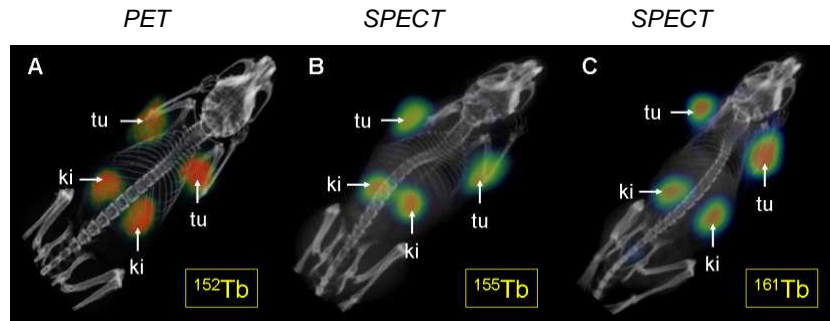


Dy 150 7.2 m	Dy 151 17 m	Dy 152 2.4 h	Dy 153 6.29 h	Dy 154 3.0 · 10 ⁴ a	Dy 155 10.0 h	Dy 156 0.056	Dy 157 6.1 h	Dy 158 0.095	Dy 159 144.4 d	Dy 160 2.329	Dy 161 18.889	Dy 162 25.475
Tb 149 4.2 m 4.1 h	Tb 150 5.8 m 3.67 h	Tb 151 25 s 12.8 h	Tb 152 4.2 m 17.5 h	Tb 153 2.34 d	Tb 154 23 h 6.9 h 21	Tb 155 5.32 d	Tb 156 437 s 5.4 d	Tb 157 99 a	Tb 158 10.5 a 180 a	Tb 159 100	Tb 160 72.3 d	Tb 161 6.90 d
Gd 149 74.6 a	Gd 149 9.28 d	Gd 150 1.8 · 10 ⁴ a	Gd 151 120 d	Gd 152 0.20 1.1 · 10 ⁴ a	Gd 153 239.47 d	Gd 154 2.18	Gd 155 14.80	Gd 156 20.47	Gd 157 15.65	Gd 158 24.84	Gd 159 18.48 h	Gd 160 21.86

General Purpose Separator (GPS)



Theranostics with terbium isotopes



¹⁵²Tb-folate: 9 MBq
Scan Start: 24 h p.i.
Scan Time: 4 h

¹⁵⁵Tb-folate: 4 MBq
Scan Start: 24 h p.i.
Scan Time: 1 h

¹⁶¹Tb-folate: 30 MBq
Scan Start: 24 h p.i.
Scan Time: 20 min



ISOLDE

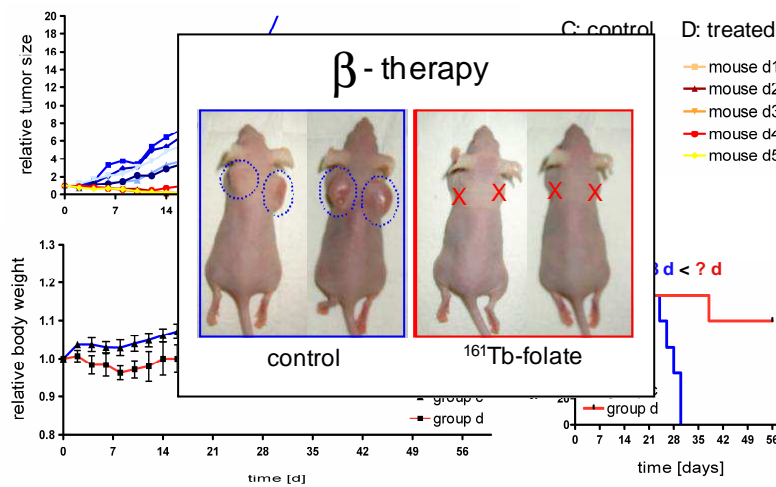


ISOLDE

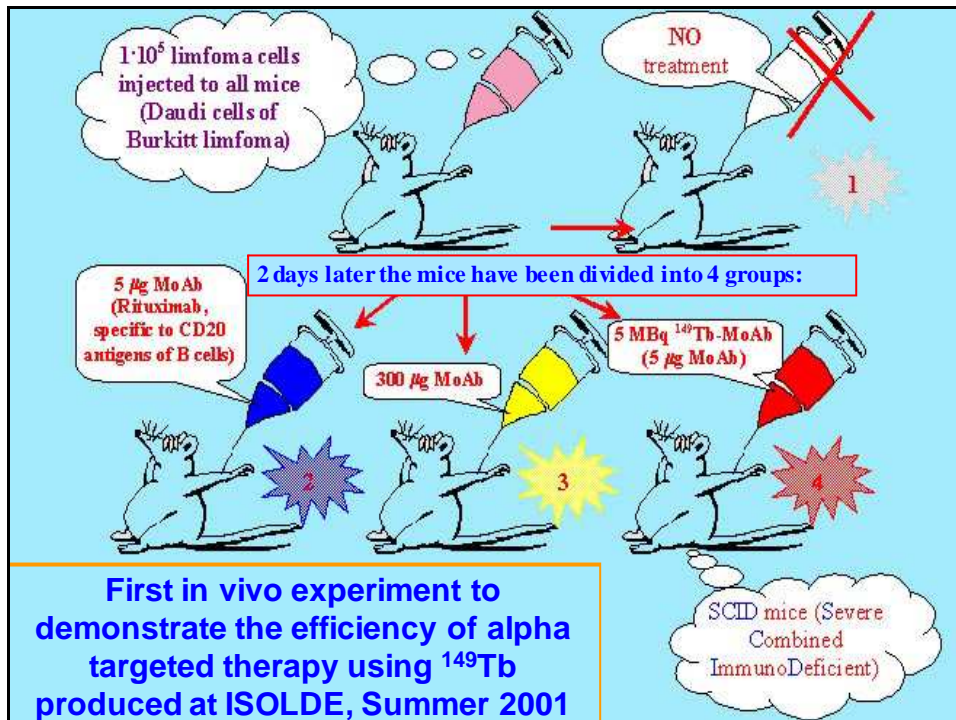
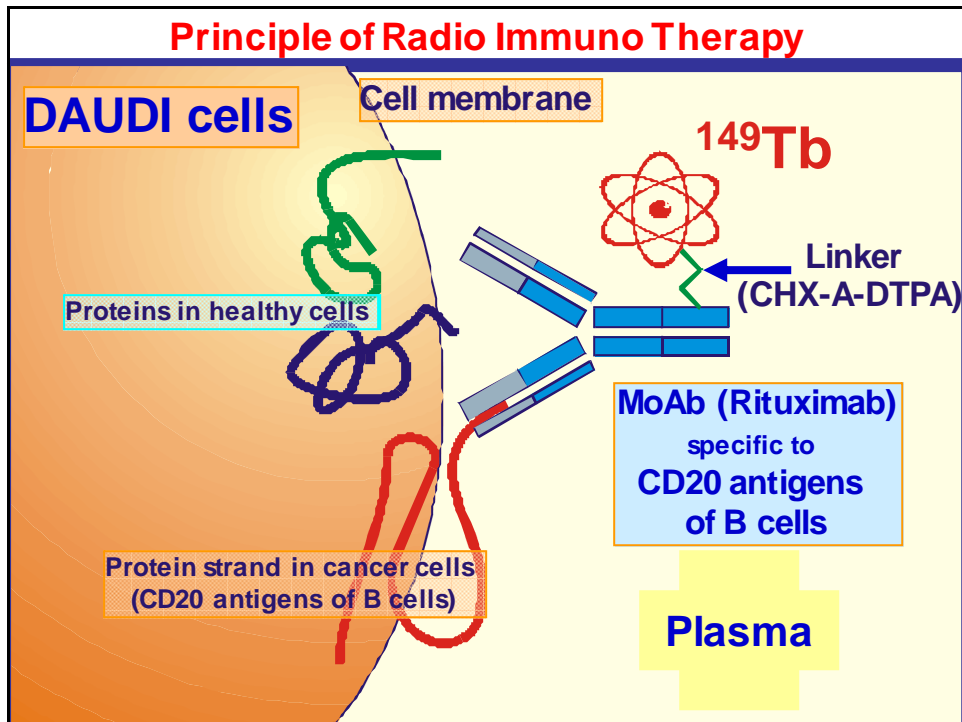


IS528 Collaboration: C. Müller et al., J. Nucl. Med. 53 (2012) 1951.

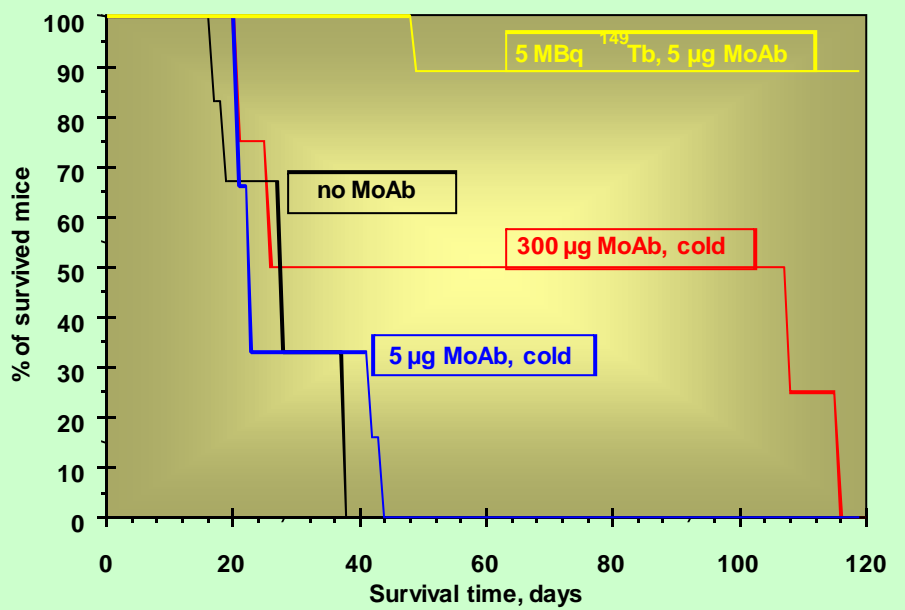
Targeted Beta Radionuclide Therapy KB Tumor-Bearing Mice Treated with ¹⁶¹Tb-Folate



IS528 Collaboration: C. Müller et al., J. Nucl. Med. 53 (2012) 1951.



^{149}Tb -RITUXIMAB in lymphoma mouse model



G.J. Beyer et al., *Eur. J. Nucl. Med. Molec. Imaging* 31 (2004) 547.